Two Flavonoid Compounds Isolated From Nepeta septemcrenata Growing in South Sinai, Egypt.

Abd El-Raouf A. Moustafa^{*1}, Hamdy Abd El-Azeem², Mohamed A. Omran³, Sawsan A.M. Nasr⁴, Ismail M. Abdel Nabi³ and Zakaria A. Teleb⁴

¹Botany Department, Faculty of Science, Suez Canal University, Egypt ²Chemistry Department, Faculty of Science, Suez Canal University, Egypt ³Zoology Department faculty of Science, Suez Canal University, Egypt ⁴Biochemistry and Molecular Biology National Organization for Drug Control and Research

*Corresponding author e-mail: raoufmoustafa2@hotmail.com

ABSTRACT

Nepeta septemcrenata is a perennial herb grows in Sinai Mountains and used in folk medicine as a laxative antispasmodic, carminative, diuretic and to treat liver diseases. The present work aimed to study phytochemical, biological and toxicological effects of Nepeta septemcrenata. The study includes phytochemical screening of N. septemcrenata and investigation of volatile oil and flavonoids of the plant. The results of preliminary studies showed that the aerial parts of N. septemcrenata contain volatile oil, carbohydrates, flavonoid and unsaturated sterols or triterpenoids while the stem contains carenolides, tannins, saponins, alkaloids and coumarins. The analysis of volatile oil of the plant (petroleum ether fraction) using GC/MS revealed that it contains 15 compounds, the main components of it was ester which represent 18.15% from the volatile oil contents. Methylene chloride fraction subjected to preparative TLC that led to the isolation of two flavonoid compounds Ne2, Ne4. Ethyle acetate fraction was subjected to a chromatographic polyamide column. The residue obtains from fractions showed by TLC and PC contained a mixture of two flavonoid compounds (Ne2 & Ne4). It was further purified by preparative paper chromatography. The two isolated compounds (Ne2 and Ne4) were identified using different techniques as preliminary chemical test, thin chromatographic examination with or without treatment with ammonia and aluminum chloride, UV spectral analysis and mass spectroscopy. The isolated compounds were identified as apigenin -7- methyl ether and flavones aglycone.

Keywords- Labiatae, South sinai, Medicinal species, Folk medicine, Flavonoids.

INTRODUCTION

Medicinal plants have a very important role from the medical and economical point of view all over the world, where the folk medicine made a wide base used for searching new drug. The Egyptian desert has many wild medicinal plants which have a great economic and therapeutic value. Family Lamiaceae comprises many species of special economic and medicinal importance due to their different constituents¹. It is represented in Egypt by 21 genera: Thymus, Marrbium, Nepeta, Mentha. Otostegia, Salvia. Ocimum. Teucrium, Orthosiphon, Lavandula. Origanum, Satureja, Micromeria, Ziziphora, Stachys, Lamium, Ballota, Leuces, Phlomis, *Prasium* and $Aiuga^2$. Several species from lamiaceae are used in folk medicine. The pharmacological activities of these plants anti-inflammatory, are antiallergic, antitumor. antimicrobial and specific therapeutic activity on gastrointestinal and respiratory tracts³. Plants belonging to the family lamiaceae are characterized by their volatile oil content. On this basis, the family is divided into two subfamilies; one of them rich in its volatile oil content is (Nepetoideae) such as Nepeta, and Rosmary, While the other is poor in its volatile oil content (Lamioideae) such as Otostegia and *Teucrium*⁴.

Nepeta is one of the most important genera of lamiaceae family with regard to the number of its species. Some species of this genus are important medicinal plants and their extracts have been used for medicinal purposes⁵. Some species of *Nepeta* germs are utilized in folk medicine for treatment of contusions, rheumatic pains, fever, cutaneous eruption and some species are used for their anti-inflammatory properties⁶. *Nepeta* plants were prepared as tea and used in traditional medicine as anthelmintics, febrifuges, expectorants and to treat bronchitis, bites as well as stings of scorpions². They have antispasmodic, astringent properties⁷, cardiotonic (*N. ciliaris*) diaphoretic, and anti-catarrhal effects (*N. hederacea*)⁸. The genus *Nepeta* is characterized by the presence of volatile oils, terpenes, flavonoids and phenolic acid. There is only one species found in Egypt (*Nepeta septemcrenata*)².

Genus Nepeta, family lamiaceae, includes 250 species of perennial plants. They are aromatic plants widely grown in Mediterranean countries⁹. The genus Nepeta (lamiaceae) is a wide spread weed in Egypt¹⁰. Nepeta septemcrenata Eherenb is an erect slender plant with branches at base leaves are oppositely alternated, ovate with crenate or slightly dentate margins. This plant is found in Saint Catherine, South Sinai, Egypt⁹. Nepeta septemcrenata known to be used by the native Bedouins in folk medicine as antipyretic, sedative. cardiotonic, eye wash and as a gargle in sore throat³. The present study was undertaken to investigate the following: A phytochemical analysis of N. septemcrenata plant includes; phytochemical screening, investigation of some chemical constituents that could be present and evaluation of the biological and therapeutic activities of the ethanolic extract of the plant including determination of its LD₅₀, anti-Inflammatory, antipyretic and the antimicrobial effect.

MATERIALS AND METHODS

Plant collection and Preparation of plant extract

The fresh *Nepeta septemcrenata* was collected from Saint Catherine mountains in South Sinai (Egypt) during the flowering season. Following¹¹ method the extract was prepared, where 440 gm of the air dried powder of the aerial parts (leaves and flowers) of *Nepeta septemcrenata* were defatted with petroleum ether (40-60 ^oC) in a soxhlet apparatus. The defatted powder

was then extracted with 95% alcohol; the alcohol extract was evaporated to dryness under reduced pressure.

Phytochemical analysis

The powdered samples of the aerial parts were subjected to chemical screening according to the standard procedures for the following constituents; volatile oil¹², carbo-hydrates and/or Glycosides¹³, alkaloids¹⁴ and/or nitrogenous bases, cardinolides¹⁵, anthraquinon¹⁶, tannins¹⁷⁻¹⁹, flavonoids²⁰, saponins²¹, sterols and/or triterpenes¹⁷, lactones and/or esters and cumarines²².

The isolation of volatile oil and extraction of flavonoid from leaves were performed according to the method of^{23} and¹¹. The survey of flavonoid was employed using paper chromatography (Whatman1), tertiary butanol: 15% acetic acid: Water (3:1:1) were used as solvents. The different fractions were also subjected to paper chromatography (PC) using 30% acetic acid as a solvent. Flavonoid profiles were viewed under UV- light and colors were recorded before and after fuming with ammonia vapour. Compounds were indentified using standard UV- visible spectroscopy (Beckmam-26 spectrophotometer), Mass Spectrophtometer (finnigan SSQ7000 model) and Gas Chromatography (Thermo- Trace 2000 model), and thin layer chromatography (TLC) recoated silica gel 60F254 plates (Merck) and polyamide plates (Merck).

Identification of the volatile constituents

The volatile constituents were prepared from the fresh aerial parts by hydro-distillations. The volatile constituents were extracted with petroleum ether (40-60 °C) and dried over anhydrous sodium sulphate¹². The prepared volatile constituents analyzed were by gas chromatography (GC/MS).

Identification of the volatile constituents was achieved by library

searched data base Willey 229 LIB and by comparing their mass fragmentation patterns with those of the available published data²⁴.

Separation of the flavonoid

After extraction with petroleum ether (40-60°c), about 440 gm of dried powder of the plant material (leaves and flowers) was extracted with ethanol till exhaustion. The ethanolic extract was concentrated and the residue was dissolved in hot distilled water. The aqueous solution was extracted by shaking with successive portion of methylene chloride $(4 \times 100 \text{ ml})$. The methylene chloride extract (fraction A) was dried over anhydrous sodium sulphate then concentrated to small volume. The aqueous mother liquor was extracted by shaking with successive portions of ethyl acetata (4×100) extract (fraction B), then it was dried over anhydrous sodium sulphate and concentrated to a small volume¹¹.

Thin layer chromatography of the contained fractions was carried out using silica gel G, and polyamide pates 0.25 mm thick coated plates. Detection of the flavonoide constituents was done by examining the chromatoplates under the U.V light at 366 nm, as by spraying with aluminum chloride before and after exposure to ammonia vapour. Whereas fluorescence spots of the flavonoid component appeared with varying colors and fluorescence¹¹. Plates coated with silica gel G and developed with chloroform-Methanol (98:2) gave the best resolution (Scheme 1).

Investigation of methylene chloride fraction

The methylene chloride fraction subjected to preparative thick layer chromatography on silica gel (SG) (0.5 mm thick) coated plates. The methanolic solution of methylene chloride fraction was applied to the plates as homogenous bands. The chromatoplates were developed with chloroform: methanol (98:2) as a solvent then the plates were dried by air. The separation bands were localized in the UV-light, scrapped off, and transferred into flasks then eluted with methanol, then filtered. The solution of the eluted bands was evaporated until dryness and the residue was dissolved in methanol¹¹.

Investigation of ethyl acetate fraction

Ethyl acetate fraction was subjected to a chromatographic polyamide column. About 400 mg of ethyl acetate fraction was applied on the top of the column. Elution was made with 70% methanol and 100 ml fraction was collected. The chromatographic separation was followed by the thin layer chromatography on SG plates. The residue obtained from fraction number (20-34) showed by TLC and PC contained a mixture of two flavonoid compounds. It was further purified by preparation paper chromatography on 3 MM paper which developed in 30% acetic acid as a solvent, then dried by air. The separation bands were localized in the UV-light¹¹.

RESULTS

Phytochemical screening of Nepeta septemcrenata

Table (1) showed the preliminary screening of the aerial parts "flowers, leaves and stem" of *Nepeta septemcrenata*. It is clear from the table that, aerial parts of *Nepeta septemcrenata* containing a highly concentration of volatile oils, carbohydrate, flavonoides and unsaturated sterols. In the mean time the other constituents "cardenolides, tannins, saponins, alkaloids and coumarins" were found in trace amount.

Volatile oils

The yielded oil of *Nepeta septemcrenata* was yellow color and has a characteristic odor. Table (2) revealed the analysis of the volatile oil of *Nepeta septemcrenata* by GC/MC. These results

indicated the presence of fifteen compounds which represent 60.82% of the total oil. The identified compounds represent several classes; ester 21.5% which chemical represent the main constituent (no. 1, 3, 5, 12 & 15 in the table and their mass illustrated in figure 1 (a),(c) and figure 2 (i) respectively), alcohol 10.23 % (no. 6, 7, & and 10 in the table and their mass illustrated in figure 1 (e), (f) and (g)), keton 6.31%(no. 11 in the table and its mass illustrated in figure 1 (h)) hydrocarbons 5.79 % (no. 2, 13, & 14 in the table and illustrated their mass in figures1 (b) and figure 2 (k), (l)) and phenol 0.95 % (no. 4 in the table and its mass illustrated in Figure 1 (d)).

The following figures from (a-h) represent the mass spectrum of the 8 compounds of the volatile oils.

The following figures from (i-l) represent the mass spectrum of the compounds 4 of the volatile oils.

Isolation and Identification of Flavonoid Compounds

As we describe before in the material and methods section, two flavonoides components, Ne2 and Ne4, were isolated from methylene chloride. TLC and PC of the isolated flavonoids showed to be single compound, identified by UV and MS spectral analysis as apigenin 7-methyl ether and flavones glycoside. Also the residue which obtained from the fraction B of the ethyl acetate extract showed by TLC and PC to be a mixture of two flavonoides compounds and the methanolic solution of the eluted bands gave upon concentration of yellow crystalline compounds Ne2 and Ne4 as the same in methylene chloride fraction.

The flavonoid substance Ne2 showed a single spot when purified by sephadex LH-20 column (pure substance), thin layer and paper chromatogram. The ultra-violet absorption spectra of the flavonoid substance Ne2 in methanol as well as in the presence of shift reagent were measured,

and the results obtained are shown in table (3). The results obtained as shown in table (3) are in agreement with these reported for apigenin -7- methyl ether. The ultra-violet spectrum in methanol showed that the flavonoid substance belong to the flavones type free hydroxyl group at C-5 was proved by the bathochromic shift in B-I-(54 nm) with ALCL₃ no free hydroxyl group at C-7 was confirmed by sodium Acetate (No shift was observed). of B-II The large bathchromic shift of b-i with sodium methoxide, is indicating of a free hydroxyl group at C-4. The mass spectrum showed a molecular ion peak (M^+) at m/z 280 (100%) which corresponds to the molecular formula $(C_{16}H_{14}O_5)$. The fragmentation pathway of the molecular ion (scheme 2) Undergoes the retro- Diets- Alder reaction, giving ring- A fragment at m/z 166 (6%) and ring - C fragment at m/z 118 (4%), fragment ion peak at obtained data the identity of the flavonoid substance Ne2 was proved to be apigenin-7- methyl ether.

The flavonoid substance Ne4 showed a single spot when purified by Sephadex LH20 column (pure substance), on thin layer and paper chromatogram. The ultra-vilot absorption spectra of shift reagent were measured, and the results obtained are shown in table (4). The UV- spectrum showed that the flavonoid substance belong to the flavone type. Mass spectrum showed that the compound belong to the flavone glycoside.

DISCUSSION

Family lamiaceae have many medical plants, which characterized by the presence of volatile oils, flavonoids, phenolic acids, terpens and coumarins. The studied plant *Nepeta septemcrenata* is belonging to the family lamiaceaa. *N. septemcrenata* is common plant in Sinai and Saint Cathrine, Egypt⁹. An isopimarance type diterpence and 7-omethyl apigenin

were isolated from the ethanolic extract of septemcrenata Ehrenb herb²⁵. Nepeta Investigated the chemical constituents of Nepeta septemcrenata²⁶ analyzed the volatile oil of the same plant species and found that it is used in folk medicine by the native Bedouins as sedative, antiseptic, antiplasmodic, antipyretic cardiotonic, eye wash and as a gargle in sore throat. The present study was aimed to investigate the phytochemical analysis of the plant, evaluate the biological and therapeutic activities of the ethanolic extract of the plant and to study the biochemical effect of the plant extract.

In the present study petroleum ether extract of Nepeta septemcrenata yielded volatile oil with yellow color and characteristic odor. The results of GC/MS analysis of this oil showed that it is contain a mixture of 15 compounds represent several chemical classes. The obtained results are in agreement with²⁶ who detected sixteen major volatile compounds of Nepeta septemcrenata and with³ who mention that *N. septemcrenata* is rich by volatile oil. In this respect ⁴ divided family lamiaceae into two subfamilies on the basis of their volatile oil content. The lamiodeae is poor in its volatile oil content while the subfamily neptoideae is rich in its volatile oils content *N. septemcrenata* belongs to subfamily Neptoideae.

In the present study, two flavonoids have been isolated from *N. septemcrenata*, the first one was apigenin-7-methyl ether. This result is in agreement with that of²⁵ who isolated and indentified 7-O-methy apigenin from the ethanol extract of the aerial parts of *N. septemcrenata*. The second flavonoids were a flavones glycoside. Identification of the isolated flvonoids was achieved through chromatographic studies and spectroscopic measurements (UV and MS/GC).

Administration of the ethanolic extract of *Nepeta septemcrenata* to the

hyperthermeric rats decreased the body temperature significantly. This result is in agreement with that of^{27} who reported that methanol extract of *Leucas lavandulaefolia* (lamiaceae) produced significant dosedependent lowering of body temperature in yeast-provoked elevation of body temperature in rats.

The interaperiotenal injection of ethanolic extract of Nepeta septemcrenata into rats with inflamed paw significantly decreased the thickness of the inflamed rats' paws. This anti-inflammatory effect was reported by⁶ who studied the antiinflammatory activity of Nepeta sibthorpii methanol extract and they observed a significant inhibition of paw oedema. They suppose that the ant-inflammatory effect of methanol extract could be related to free radical scavenging activity and it depends on a synergic effect of component of the alcoholic extract. So, it appears that the antinflammatory action of the alcoholic extract of N. septemcrenata could be due to its antioxidant properties.

It was found that apigenin as a flavonoid of Nepeta setemcrenata extract inhibit the induction of β -galactosidase activity in Sw480 cells stably transfected β-galactosidase. Apigenin with demonstrated anti-inflammatory potent activity in carrageenan induced rat paw edema and it was concluded that flavonoids offer important therapeutic potential for the treatment of a variety of inflammatory diseases²⁸. It was investigated that the bioflavonoids on lysosomal acid hydrolylases, viz., β -glucuronidase, β -Nacetyl glucosaminidase and cathepsinn D in serum, liver, kidney and spleen and the stability of liver lysosomes was studied by 29 . The activity of these enzymes in arthritic tissue and serum increased significantly. The total liver was appreciably decreased, while its release was significantly. The total activity of β -glucurandase in the lysosomerich fraction from arthritic liver was

appreciably decreased, while its release was significantly increased.

It was found that one of the most potent flavones, apigenin as a flavonoid exhibited a dose and time dependent, reversible effect on adhesion protein expression as well as inhibitory adhesion protein up regulation at transcriptional level²⁸. Also, apigenin inhibited IL-1 α induced prostaglandin synthesis and TNF- α -INDUCED 1L-6 and 1L-8 productions, suggesting that the hydroxyl flavones may acts as general inhibitors of cytokineinduced gene expression. As well as, this flavonoid did inhibit TNF- α -induced β glactosidase activity in SW 480 cells stably tranfected will a β - glactosidase.

It was found that the natural extract of some medical plants protector free radicals damage³⁰. In addition, it has been shown that the content of the mucosal non protein. Sulfhydrxyl group may increase by the effect of the active ingredinents of the medicinal plants³¹.

It was found that the protective effects of four flavonoids "quercetin, rutin, luteolin and apigenin" against induced DNA damage in human leukemia cells (K562). These flavonoids are characterized by the number of hydroxyl groups on the B-ring, the presence of an O-dihydroxy structure on the B-ring confers a higher degree of stability to the flavonoid phenoxyl radicals by participating in electron delocalization and is, therefore, an important determinant for antioxidant potential³². The strongest inhibition against DNA damage (44%, 42%) was found in a range of luteolin and quercetin concentrations of 20-100 u mol/L.

It has been investigated that DNA protective capacity of three flavonoids, apigenin, luteolin and quercetin against the free radical generated by H_2O_2 . The quantitive analysis has shown that luteolin possesses the highest DNA protective effect of flavonoids³³. Also, it has been found that thirteen isoflavonoids, flavonoids, and

lignans were evaluated for their effects on DNA synthesis in MCF-7 and human breast cancer cells by³⁴. At 0.1-10 uM of coumestrol, genistein, biochanin A, luteolin, apigenin kaempferol and enterolactone induced DNA synthesis 150-235% and at 20-90 uM, inhibited DNA synthesis by 50%. The biphasic effects showing inhibition of DNA synthesis at high concentrations but induction at concentration close to probable levels in humans³⁵. Investigated that the "aurestin. flavonoides rhamnetin. isorhamnetin, apigenin, and luteolin" did not introduce any damage into the DNA.

In conclusion, *N. septemcrenata* has many promising active constituents that exert significant pharmacological and therapeutic effects with high margine of safety. We recommend further phytochemical and pharmacological studies on each purified active constituent of this plant on different body systems and on fatal diseases such as cancer and AIDS.

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Table 1. Preliminary screening of the aerial parts (flower, Leaves and stem) of Nepeta septemcrenata

Constituents	Leaves and flowers	Stem
Volatile oil	+++	+++
carbohydrate and /or glycosides: (Molisch`s test)	+++	+++
Test for cyanogentic glycosides (Guignard reaction)	-	-
Cardenolides	-	+
Anthraquinone	-	-
Free anthraquinone aglycones	-	-
Combined aglycnes	-	-
Tannins	-	+
Flavonoids:	+++	+++
Saponins (Forth test)	-	+
Unsaturated sterols and/or triterpenoids	+++	+++
Alkaloids or nitrogenous base	-	+
Lacto and / or ester	-	-
Coumarins	-	+

+++High concentration, ++Moderate concentration, + Trace concentration, -Absent

Table 2. GC/MS analysis of the volatile oil of Nepeta septemcrenata

Compound	R.T.	%	M.F.	B.P.	M *
2-Nonenoic acid, melthyl ester, (E)	24.07	6.10	$C_{10} H_{16} O_2$	87	170
Cyclopropapane, (2- methyylenebutyl)	24.78	2.21	C ₈ H ₁₄	81.16	110
4-Nenenoic acid, methyl ester	27.04	1.32	$C_{10} H_{18} O_2$	81.16	110
Phenol, bis (1,1- dimethylethyl)	30.73	0.95	$C_{14} H_{22} O$	59.13	206
Butanoic acid, 3-methyl ester	30.96	0.58	$C_6 H_{12} O_2$	59.16	117
1-Hexadecanol	32.25	5.88	C ₁₆ H ₃₄ O	83.16	242
2,2,6-trimethyl-7,9-di(methylidine) bicycle(4,3.0) nonan-1- ol	33.49	0.95	C ₁₄ H ₂₂ O	91.04	206
Bisabolol oxide A	35.82	7.95	$C_{15} H_{26}O_2$	59.14	238
Germacrane-D	35.82	8.10	$C_{15} H_{30}$	97.11	210
3,5 –O- Benzylidene-1,2-dideoxy-D-erythro-pent-1-ynitol	37.37	3.36	$C_{14} H_{14} O_4$	105.11	246
Benzene, (1- butynony1)	37.75	6.31	$C_{19} H_{32}$	91.09	260
®-2-(1`,5`- Dimethylhex-4`-Enyl)-5-Methylphenylacetate	38.19	3.43	$C_{17} H_{24} O_2$	91.08	260
1-ethylundecylbenzene	38.66	1.13	$C_{19} H_{32}$	91.11	260
1-Methyldodecylbenzene	39.59	2.45	C ₁₉ H ₃₂	105.10	260
2-pyridinepropanoic,a`-methyl-a`-oxo-ethyl ester	61.89	10.15	$C_{11} H_{13} NO_3$	73.12	207

R.T.: Retention time, M.F.: Molecular formula, M⁺: Molecular weight, B.P.: Base peak.

Band	Max (n.m.)					
	MeOH	NaOMe	AICI₃	AICl ₃ / HCl	NaOAc	$NaOAc/H_3BO_3$
Ш	267	266	270	243	266	267
I	329	380	383,346	379,339	382	332

 Table 3. Ultra- violet absorption spectra of Ne2 in methanol

Table 4. Ultra- violet absorption spectra of shift reagent of Ne4

Band	Max (n.m.)					
	MeOH	NaOMe	AICI ₃	AICI ₃ / HCI	NaOAc	NaOAc/H ₃ BO ₃
П	267	258,247	272	275	269	242
I	398	398	390	355	438	348

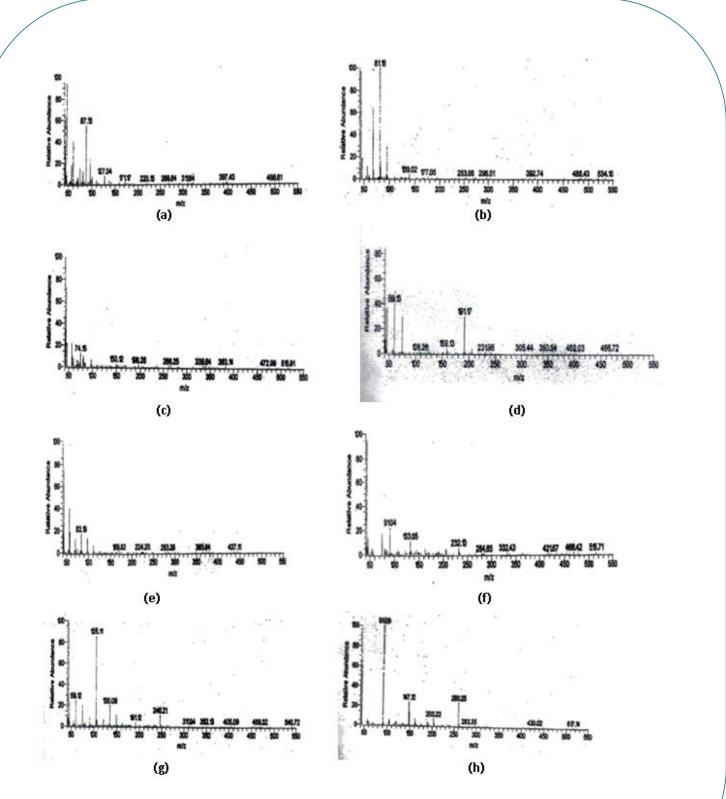
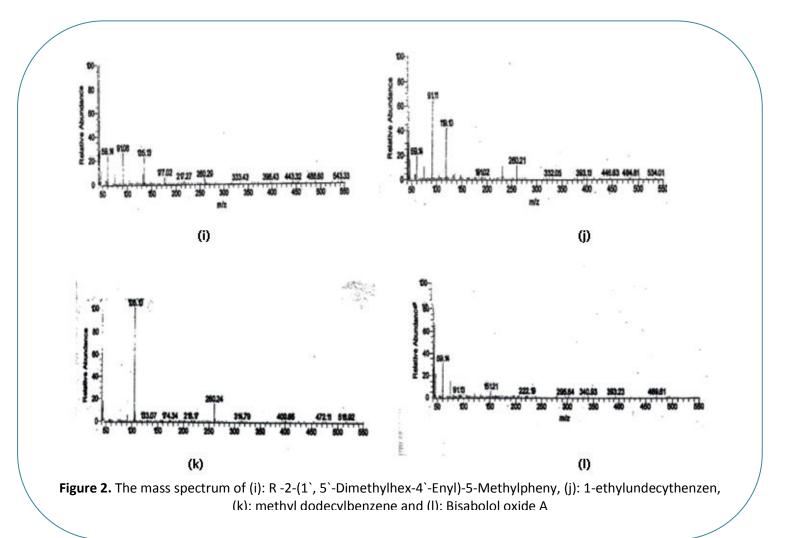
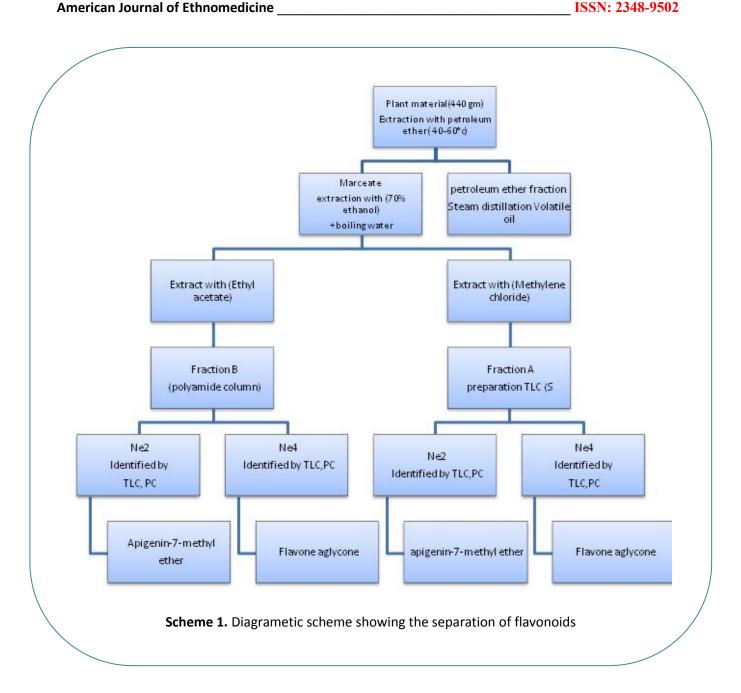


Figure 1. The mass spectrum of (a): 2-Nonenoic acid, methyl ester,(E), (b): Cyclopropane, (2-methylenebutyl), (c): 4-Nenenoic acid, methyl ester, (d): Phenol, bis (1,1-dimethyethy), (e): 1-Hexadecanol, (f): 2,2,6-trimety11-7,9-di (metyhylidine) bicycle (4,3.0) nonan-ol, (g): 3,5-O-Benzyidene-1,2-diddeoxy-D-erthro-pent-1-ynitol and (h): Benzene, (1-butylnony)







American Journal of Ethnomedicine

