# Structural and conformational studies of 2,4-diphenyl-3-azabicyclo[3.3.1]nonan-9-one-o-nicotinoyloxime 

T. Mohandas ${ }^{\text {a }}$, K. Gokula Krishnan ${ }^{\text {b }}$, V. Thanikachalam ${ }^{\text {b }}$ and P. Sakthivel ${ }^{\text {c }}$<br>${ }^{a}$ Department of Physics, Shri Angalamman College of Engineering and Technology, Tiruchirappalli, India<br>${ }^{b}$ Department of Chemistry, Annamalai University, Chidambaram, India<br>${ }^{c}$ Department of Physics, Urumu Dhanalakshmi College, Tiruchirappalli, India


#### Abstract

The title compound crystallizes with the monoclinic cell setting and P2(1)/n space group. The structural compound $\mathrm{C}_{26} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{2}$ exist in a twin-chair conformation with equatorial orientation of the phenyl group. The two phenyl rings attached to the piperidine ring at positions 2 and 4 have equatorial orientation. The angle between the two benzene rings is $48.32(9)^{\circ}$. The crystal structure is further stabilized by C-H...N, C-H... pi and intramolecular interaction.


Key words: Piperidone, oxime, hydrogen bonding

## INTRODUCTION

Oxime esters are serving as important synthetic intermediate, and have been employed as starting materials for both synthetic and medicinal chemistry [1a,b,c]. Tropane, cocaine and granatane (pesudopelletierine) are some azabicyclo types of alkaloids found in pharmacologically and medicinally active drugs[2]. The azabicyclo [3.3.1]nonane is a serotonin $5-\mathrm{HT}_{3}$ receptor[3], and also used as anti-Parkinson's[4], local anesthetic[2]. By using the X-ray diffraction method the structure of the above compound is to be discussed. The molecular structure of the title compound is shown in fig.1.


Figure 1

## MATERIALS AND METHODS

A mixture of 2,4-diphenyl-3-azabicyclo[3.3.1]nonan-9-one oxime ( $0.765 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) and nicotinic acid ( 0.338 g , 2.75 mmol ) in dry pyridine ( 5 mL ) was stirred at ambient temperature. $\mathrm{POCl}_{3}(0.25 \mathrm{~mL}, 2.75 \mathrm{mmol})$ was added drop wise to the reaction mixture and stirring is continued for 40 to 45 min . The progress of the reaction was monitored
by TLC. After completion of the reaction, a saturated solution of $\mathrm{NaHCO}_{3}$ was added portion wise to the reaction mixture and the crude product was thrown out as a precipitate. The crude product was then recrystallized from absolute ethanol to get the pure 2,4-diphenyl-3-azabicyclo[3.3.1]nonan-9-one $O$-nicotinoyl oxime. Yield 0.64 g ( $78 \%$ ).The crystal was developed on slow evaporation techniques using absolute ethanol as a solvent.

## Intensity Data Collection

X-ray data were collected on a Bruker AXS (Kappa APEX2)[5] CCD area detector using $\omega$ and $\psi$ scan mode. A small crystal of size $0.30 \times 0.25 \times 0.23 \mathrm{~mm}$ was chosen and its quality was checked using polarizing microscope. Cell refinement and data reduction were carried out by using APEX2/SAINT-NT [5]. Sets of two standard reflections were monitored for every one hour of exposure during the data collection and there was no noticeable change in the intensity observed. A total of 25944 reflections were collected resulting in 5789 independent reflections of which 3711 had $\mathrm{I}>2 \sigma(\mathrm{I})$, were considered as observed reflections. The intensities were corrected for Lorentz and polarization effects. Absorption corrections were made with SADABS[5].

## Structure Solution and Refinement

The refinement was carried out by SHELXL-97[6].The positions of the hydrogen atoms bound to the O and C atoms are identified from the difference electron density maps and their distances are geometrically optimized. The H atoms associated with the hydroxyl groups are constrained to a distance of $\mathrm{d}(\mathrm{O}---\mathrm{H})=0.82 \backslash \% \mathrm{~A}$; and $\mathrm{U}_{\text {iso }}(\mathrm{H})=$ $1.5 \mathrm{U}_{\text {eq }}(\mathrm{O})$. The hydrogen atoms bound to the C atoms are treated as riding atoms, with $\mathrm{d}(\mathrm{C}---\mathrm{H})=0.93$ and $\mathrm{U}_{\mathrm{iso}}(\mathrm{H})=$ $1.2 \mathrm{U}_{\mathrm{eq}}(\mathrm{C})$ for aromatic, $\mathrm{d}(\mathrm{C}---\mathrm{H})=0.97$ and $\mathrm{U}_{\text {iso }}(\mathrm{H})=1.2 \mathrm{Ueq}(\mathrm{C})$ for methylene and $\mathrm{d}(\mathrm{C}---\mathrm{H})=0.96$ and $\mathrm{U}_{\text {iso }}(\mathrm{H})$ $=1.5 \mathrm{U}_{\mathrm{eq}}(\mathrm{C})$ for methyl groups.

The ortep diagram is shown in fig 2.


Figure 1
The weighting scheme [13] adopted during the final cycle of refinement is
$w=1 /\left[\backslash s^{\wedge} 2^{\wedge}\left(\mathrm{Fo}^{\wedge} 2^{\wedge}\right)+(0.0603 \mathrm{P})^{\wedge} 2^{\wedge}+0.1935 \mathrm{P}\right] .$,
where $\mathrm{P}=\left(\mathrm{Fo}^{\wedge} 2^{\wedge}+2 \mathrm{Fc}^{\wedge} 2^{\wedge}\right) / 3^{\prime}$

The geometric calculations were performed using the PARST[7,8,] and PLATON [9].

## RESULTS AND DISCUSSION

In the title compound, the cyclohexanone ring deviates from an ideal chair with total puckering amplitude $\mathrm{QT}=$ $0.5633^{\circ}$. and the piperidine ring is closer to an ideal chair , $\mathrm{QT}=0.6077^{\circ}$. The piperidine ring (C7/C8/C9/C10/C11/N1 )exists in a chair conformation with the puckering amplitude $\mathrm{Q}=0.6077(15) \AA$, theta $=2.14(13)^{\circ}$, and phi $=163(4)^{\circ}$.Also the ring (C8/C9/C10/C18/C19/C20) exists in a twin chair conformation with the puckering amplitude $\mathrm{Q}=0.5638(16) \AA$, theta $=166.11(17)^{\circ}$ and $\mathrm{phi}=250.0(7)^{\circ}[10]$. The torsional angles of $\mathrm{C} 9 / \mathrm{C} 10 / \mathrm{C} 11 / \mathrm{C} 12$ and $\mathrm{C} 1 / \mathrm{C} 6 / \mathrm{C} 7 / \mathrm{C} 8$ are $177.91(11)^{\circ}$ and $-97.24(14)^{\circ}$ respectively. The piperidine ring makes the dihedral angle of $55.99(8)^{\circ}$ and $76.33(8)^{\circ}$ with the phenyl rings respectively. The angle between the two phenyl rings is $48.32(9)^{\circ}$.

The bond distances and bond angles in the title compound agree very well with the corresponding values reported in closely related compound [11,12].

The crystal structure of the compound is stabilized by $\mathrm{N}-\mathrm{H} . . . \mathrm{N}, \mathrm{C}-\mathrm{H} . . . \mathrm{N}, \mathrm{C}-\mathrm{H} . .$. pi and intramolecular interaction. where $\operatorname{cg} 2$ is the centroid of (C25/C24/C23/C22/C26/N3) and $\operatorname{cg} 5$ is the centroid of (C12/C13/C14/C15/C16/C17) respectively. In the crystal ,the intramolecular infraction form infinite one dimensional chain along (001) direction shown in Fig.


Figure 2
Table :1 Crystal data and structure refinement of the title compound


Table 2: Torsional angles of non -hydrogen atoms

| Atoms | Angles |
| :---: | :---: |
| C2 C1 C6 C7 | $179.75(13)$ |
| C4 C5 C6 C7 | $-179.53(12)$ |
| C1 C6 C7 N1 | $26.35(17)$ |
| C5 C6 C7 N1 | $-154.57(12)$ |
| C1 C6 C7 C8 | $-97.24(14)$ |
| C5 C6 C7 C8 | $81.84(14)$ |
| N1 C7 C8 C9 | $58.54(12)$ |
| C6 C7 C8 C9 | $-176.23(10)$ |
| C9 C10 C11 C12 | $177.91(11)$ |
| N1 C7 C8 C18 | $-61.92(14)$ |
| C6 C7 C8 C18 | $63.32(14)$ |
| C18 C8 C9 N2 | $-115.87(13)$ |
| C7 C8 C9 N2 | $119.10(13)$ |
| C18 C8 C9 C10 | $63.58(13)$ |
| C7 C8 C9 C10 | $-61.46(13)$ |
| N2 C9 C10 C20 | $114.47(16)$ |
| C8 C9 C10 C20 | $-64.86(13)$ |
| N2 C9 C10 C11 | $-120.41(16)$ |
| C8 C9 C10 C11 | $60.25(14)$ |
| C9 C10 C11 N1 | $-57.12(13)$ |
| C20 C10 C11 N1 | $61.48(14)$ |
| C20 C10 C11 C12 | $-63.49(15)$ |
| N1 C11 C12 C17 | $-47.38(17)$ |
| C10 C11 C12 C17 | $77.27(16)$ |
| N1 C11 C12 C13 | $131.62(13)$ |
| C10 C11 C12 C13 | $-103.73(15)$ |
| C11 C12 C13 C14 | $178.93(13)$ |
| C13 C14 C15 C16 | $1.4(3)$ |

Table3: Hydrogen bonding geometry ( $\AA \AA^{\circ}{ }^{\circ}$ ) (D-donor; A-acceptor; H-hydrogen) for compound

| D | H | A | D-H | H..A | D..A | D-H..A | A-site |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| N1 | H1A | N3 | 0.869 | 2.3030 | $3.1446(18)$ | 163.3 | $1 \_656$ |
| C7 | H7 | Cg2 | 0.980 | 2.8100 | $3.7888(15)$ | 174 | $3 \_556$ |
| C20 | H20A | Cg2 | 0.970 | 2.9500 | $3.6588(17)$ | 131 | 4_555 |
| C25 | H25 | Cg5 | 0.930 | 2.7500 | $3.5806(17)$ | 149 | $1 \_454$ |

symmetry code:(i)1+X,Y,1+Z ii)-X,-Y,1-Z iii)1/2+X,1/2-Y,1/2+Z iv) $-1+X, Y,-1+Z$

## CONCLUSION

The title compound adopts a monoclinic cell setting with the standard cell parameters and the piperidine ring adopts the twin- chair conformation. Further the crystal packing was stabilized by intramolecular interaction.

We are doing active research in the area of heterocycles which are acting as potential candidates in medicinal chemistry. The diseases such as Parkinson's disease, Alzicemers are great problems in the western countries. No effective drugs are available to cure them completely. So ongoing research for developing effective drug candidate in this area urge us to do research on heterocycles active for curing them.

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## REFERENCES

[1] (a) Crichlow G. V, Cheng K. F., Dabideen,D, Ochani M, Aljabari B, Pavlov V.A, Miller E.J, Lolis E, AlAbed Y. The Journal of Biological Chemistry, 2007, 282, 23089-23095.
(b) Hwu J.R, Yang J.R, Tsay S.C, Hsu M.H, Chen Y.C, Chou S.S.P, Tetrahedron Letters, 2008, 49, 3312-3315.
(c) Neely J.M, Rovis T., Journal of American Chemical Society. 2013, 135, 66-69.
[2] Schneider M.J , Alkaloids: Chemical and Biological Perspectives, Pelletier S.W, Ed.; Pergamon: Oxford, U.K., 1996, Vol. 10.
[3] Vernekar S. K.V, Hallaq H. Y, Clarkson G ,Thompson A. J, Silvestri L, Lummis S. C. R, Lochner M, Journal of Medicinal Chemistry 2010, 53, 2324-2328.
[4] Meshi T, Nakamura S, Sato Y,Chemical and Pharmaceutical Bulletin 1972,20, 1687-1698.
[5] Bruker,APEX2,SAINT, and SADABS. BrukerAXIS Inc.,Madison,Wisconsin,USA. 2008
[6] Sheldrick, G. M. Acta Cryst.A, 2008, 64, 112--122.
[7] Nardelli, M.. Comput. Chem. 1983b, 7, 95-98
[8] Nardelli, M. J. Appl. Cryst.1995, 28, 659.
[9] Spek, A. L. Acta Cryst D, 2009,65, 148--155.
[10] Cremer, D. \& Pople, J. A.. J. Am. Chem. Soc, 1975,97,1354--1358.
[11] Park, D. H., Ramkumar, V. \& Parthiban, P..Acta Cryst E, 2012a,68, 0524.
[12] Park, D. H., Ramkumar, V. \& Parthiban, P. Acta Cryst.E, 2012b,68, 0525.
[13] Wilson, A. J, Acta Cryst, 1976, A32, 994.

