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Optical Characteristics of Manganous Doped Cadmium Oxide Thin Films Deposited by Spray Pyrolysis Method

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ABSTRACT

The properties of cadmium oxide films are very similar to those of Mn doped CdO films. They are high resistance to mechanical and chemical degradation. In this article Mn doped CdO films have been deposited on the glass substrate and this optical characteristics are analyzed by using the spectrophotometer. Films are deposited on glass substrate at various temperatures ranged from $400^{\circ}C-600^{\circ}C$ by spray pyrolysis method.

Keywords: CdO, Mn, Spray pyrolysis, Glass substrate.

INTRODUCTION

Oxides of many metals such as Tin, Indium, Zinc, Cadmium and their alloys, can be used as TCO's possessing transparent conducting property. Most of the studied transparent conducting metal oxides are anion deficient (i.e., Oxygen deficient) and hence are always n-type conductors [1]. These films can exhibit various characteristics of metals, semiconductors and insulators with improved electrical and optical properties. So they can be used as electrodes in optoelectronic devices [2], display devices [3] and photovoltaic cells [4]. CdO films can be synthesized by various methods such as Spray Pyrolysis [5], Sputtering [6], Sol-gel spin coating [7], Activated reactive evaporation [8], Metal Organic Chemical Vapour deposition (MOCVD) [9]. The transparent conducting metal CdO is coming under n-type semiconductor with 2.3 eV to 2.5 eV in direct band gap and 1.3 eV to 1.98 eV in indirect band gap [10-12]. This CdO doped films have the non-linear optical properties [13]. There has been a lot of work on the growth and physical properties of transparent conducting oxides which has lot of increasing applications [14]. These CdO doped films are common materials for detectors, gas sensors, and photovoltaic solar cells. Mn doped CdO films by spray pyrolysis technique and investigated the effect of Mn on electrical, optical, structural and surface properties of CdO films showed that it can be used in optoelectronic applications [15]. The Mn-doped CdO nano particles having the ferromagnetic behaviour at low concentration [16].

EXPERIMENTAL

Materials and methods

Solution of Cadmium acetate (0.2 M), Manganous chloride (0.1 M) are prepared by dissolving in distilled water. The ratio of Cadmium acetate and Manganous chloride is 2:1. Initially the glass substrate is heated around 450°C. After that the prepared solution was sprayed on the substrate. Then the substrate is slowly cooled. The experimental setup consists of an indigenously designed glass spray gun. It has two inlets. One is connected to a compressed air reservoir and the other is connected to the solution reservoir. The diameter of the spray gun is 0.0425 cm. The pressure of the air is measured by a pressure gauge. The deposit layers must be adherent to the substrates and there should not peel off from the substrates under the normal conditions of stress and strain, mechanical or thermal to which the deposits are exposed during their uses. A poor adhesion of a deposit to a substrate is primarily caused by the presence of a physical barrier between the deposit and the substrate at the interface due to oil, grease, undesired surface layers over the substrate surface. Presence of point defects, lattice misfits, grain misfits are also affect the deposition of film over the substrate. The effect of these can be often being reduced or completely eliminated by appropriate annealing at a

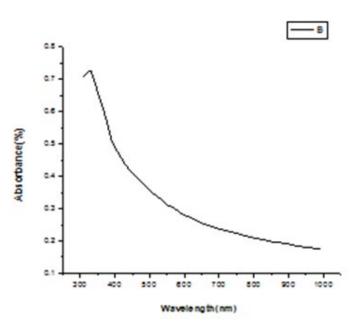


Figure 1: Absorbance (%) vs Wave length

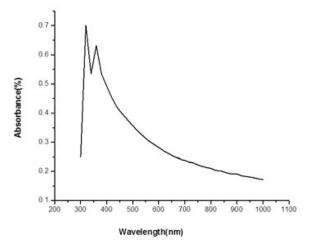


Figure 2: Absorbance (%) vs Wave length

suitable temperature or sometimes by varying the deposition conditions.

RESULTS AND DISCUSSION

The thickness of the coated film was calculated by microbalance technique by using the formula t=w/A ρ . Where 'w' is weight difference of the film, ' ρ ' is density and 'A' is deposition area. The thickness of Manganous doped CdO thin film is found to be 2.5 µm. The thickness of the thin film may varied by spraying time. The absorbance *vs* wavelength have been plotted for deposited thin film. From this graph the absorption co-efficient 'a' can be calculated for various values of wavelength. **Figure 1** shows the absorption spectra of CdO. The absorption of CdO films was studied in the wavelength range 300-1000 nm. The optical absorbance of the films decreases with the wavelength. These spectra reveal that as-grown CdO films have more absorbance in the UV region and less absorbance in the visible and high wavelength region. In **Figure 2** it is noted that the absorbance of CdO film varies between 300 to 400 nm, after that it will goes on degreasing. The band gap energy of the film on glass is determined by drawing the graph between hy *vs* (α h\gamma)3/2 and it is indicated in **Figure 3**. The corresponding band energy gap was found to be 2.773 eV. This band gap may be changed due to the difference in the deposition method and the parameters used during the process of coating.

CONCLUSION

The Manganous doped CdO thin film was grown on glass substrate by spray pyrolysis method. The thickness of the doped film was

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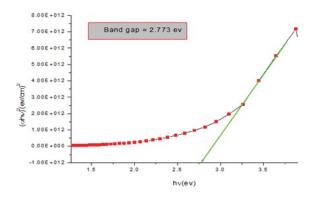


Figure 3: Determinations of band gap

calculated and it is found to be $2.5 \,\mu$ m. The absorption co-efficient 'a' can be calculated for various values of wavelength range 300-1000 nm. This shows the optical absorbance of the films decreases with the wavelength. These spectra reveal that as-grown CdO films have more absorbance in the UV region and less absorbance in the visible and high wavelength region. The corresponding band energy gap was found to be 2.773 eV. This band gap may be changed due to the difference in the deposition method and the parameters used during the process of *coating*.

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