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# Design and synthesis of novel N-substituted morpholino benzamide derivatives as antimicrobial agents

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# ABSTRACT

A mixture of 4-chloro benzonitrile and morpholine was subjected to microwave irradiation in solvent free condition to give 4-morpholino benzonitrile (1). Partial hydrolysis of (1) in 6N sodium hydroxide and 30% H<sub>2</sub>O<sub>2</sub> resulted in 4morpholino benzamide (2). A series of N-((2-hydroxynaphthalen-1-yl) (substituted phenyl) methyl)-4-morpholino benzamide **3** (**a**-**k**) was obtained in one pot synthesis by stirring **2** with aromatic aldehyde and  $\beta$ -naphthol in presence of oxalic acid as catalyst in solvent free condition. The synthesized compounds were evaluated for their anti bacterial and anti mycobacterial activity. Some of the synthesized compounds like **3b**, **3g** and **3k** have shown excellent antibacterial activity against B. subtilis and S. aureus. Amongst the compounds tested **3f** and **3h** were found to be the potent against M. tuberculosis H37Rv.

Keywords: Morpholine, Amidoalkyl naphthol, Antibacterial activity, Antitubercular activity, microwave assisted.

# INTRODUCTION

Microorganisms are exceptionally diverse, found almost everywhere and affect the human society in countless ways [1]. In particular, the emergence of multi-drug resistant strains of Gram-positive bacterial pathogens such as Methicillin-Resistant *Staphylococcus aureus* (**MRSA**) is a problem of ever-increasing significance [2, 3, 4]. Consequently, there is a vital need for the development of new antimicrobial agents having potent activity against the resistant microorganisms [5-8].

Naphthalene and its derivatives have shown a large spectrum of antimicrobial activity. Literature survey reveals that extensive research has been done on  $\beta$ -naphthol as an excellent lead moiety for designing a synthetic derivative, which posses good biologically activity [9-13]. Morpholine ring enhances the antimicrobial activity [14, 15], the marketed antibacterial drug Linezolid contains morpholine ring. In the present study, coupled synthetic derivatives containing  $\beta$ -naphthol, morpholine ring and aromatic aldehyde were synthesized and speculated to get enhanced bioactivity due to combined effect of these moieties. The development of pharmacophoric model is shown in **Figure 1**.



Figure 1 The development of pharmacophoric model for synthesized compound.

#### Chemistry

N-((2-Hydroxynaphthalen-1-yl)(substitutedphenyl)methyl)-4-morpholinobenzamide derivatives**3(a-k)**were synthesized as per the scheme of synthesis,**Scheme1.**Some of the reactions were carried out in synthetic microwave so as to get faster reaction rate and better yield.



Scheme 1 i = Solvent free, 25 min, (700 W) MW  $ii = NaOH/H_2O_2, 4h, stirring 40-50°C$  $iii = Solvent free, oxalic acid, \beta-naphthol, Ar-CHO, (8-24 min) stirring 125°C$ 

Ar	
a. 4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	g. $4$ -ClC <sub>6</sub> H <sub>4</sub>
b. 4-OH C <sub>6</sub> H <sub>4</sub>	h. 2,4-diCl $C_6H_3$
c. 4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	i. NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>
d. furfuryl	j. 4-ph
e. 2-SH C <sub>6</sub> H <sub>4</sub>	k. 4-FC <sub>6</sub> H <sub>4</sub>
f. 2-OH C <sub>6</sub> H <sub>4</sub>	

4-Chlorobenzonitrile was reacted with morpholine to give 4-morpholinobenzonitrile 1 [15]. The reaction was carried out by two methods I) Conventional, II) Microwave-assisted. The comparison of the data obtained by these methods is given in Table 1.

Mologylan formula Mologylan weight		% Yield		Time required		M D <sup>0</sup> C	D volue
Wiolecular for mula	Molecular weight	Ι	II	Ι	II	M. F. C	K <sub>f</sub> value
$C_{11}H_{12}ON_2$	188	53.28	65.33	12 h	25 min	82-83 <sup>0</sup> C	0.33

Table 1: Physical characterization data for 4-morpholino benzonitrile

Table 2: Physical constants data for synthesized derivatives 3(a-k)
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		Method I		Method II				
Sr. no.	R	%Yield	Time In hr	%Yield	Time In min	Melting Point ( <sup>0</sup> C)	R <sub>f</sub> Value	
3a		60.1	14	85.4	20	185-187	0.75	
3b	HO	65.2	13	88.9	24	160-162	0.55	
3c		71.3	16	80.1	8	215-217	0.62	
3d		59.2	16	82.4	16	141-143	0.42	
3e	SH	64.2	17	80.4	17	195-197	0.58	
3f	ОН	70.1	20	84.1	20	132-134	0.78	
3g	CI	65.3	10	91.1	10	170-172	0.52	
3h	CI	58.6	9	89.4	9	150-152	0.70	
3i		74.5	11	90.7	18	230-232	0.69	
3ј		63.5	10	90.1	8	245-247	0.83	
3k	F	73.3	11	94.1	8	180-182	0.64	

Solvent system chosen for  $R_f$  value determination was n-hexane : ethyl acetate (4 :1)

Method I= Stirring at room temperature with solvent

Method II= Stirring at  $125^{\circ}C$  without solvent

The intermediate compound 4-morpholinobenzamide 2 obtained by partial hydrolysis of 4-morpholino benzonitrile using H<sub>2</sub>O<sub>2</sub>, NaOH [16]. The final compound N-((2-hydroxynaphthalen-1-yl)(substituted phenyl) methyl)-4morpholinobenzamide 3(a-k) were synthesized by stirring at 125°C for 8-24 min without solvent in presence of catalytic amount of oxalic acid as described in general procedure [17]. All the compounds were identified by spectral data. In general, IR spectra in cm<sup>-1</sup> of 4-morpholino benzonitrile (1) showed bands at 2226 (C-N nitrile), 3091(C-C aromatic ring) 2900 (C-H alicyclic) 1190 (C-N amine). The <sup>1</sup>H-NMR showed signals at δ3.18, δ3.65 (t, 2H, CH<sub>2</sub> morpholine ring) and  $\delta$ 7.46,  $\delta$ 6.94 (d, 2H, CH<sub>2</sub> benzene) MS m/z: 188 (M). The compound morpholino benzamide (2) exhibited bands at 3363-3182 (NH<sub>2</sub> amine), 3000 (C-H aromatic), 2900 (C-H aliphatic), 1657 (C=O amide), 1408 (C-N amide). The <sup>1</sup>H-NMR presented signals at  $\delta$ 3.18,  $\delta$ 3.65 (t, 2H, CH<sub>2</sub> morpholine ring),  $\delta$ 6.94,  $\delta$ 7.60 (d, 2H, CH<sub>2</sub> benzene),  $\delta$ 7.50 (s, 2H, -CONH<sub>2</sub>, D2O exchangeable). MS *m/z*: 207 (M). The assignments of the synthesized were based on elemental and spectral data. Physical characterization data of the synthesized derivatives is given in **Table 2**.

#### MATERIALS AND METHODS

All reagents and solvents were used as obtained from the supplier or recrystallized/redistilled unless otherwise noted. Synthetic microwave oven Cata R, was used for the first step of synthesis. Melting points were determined by open capillary method and are uncorrected. Infrared spectra were recorded on JASCO FT IR (PS-4000) using KBr powder technique and frequencies are expressed in cm-1. Mass spectra were recorded on Micromasss Q-Tof Micro system mass spectrometer. 1H-NMR spectra were recorded on Varian Mercury YH 300 FT-NMR Spectrometer operating at 300 MHz (1H) and on BRUKER AVANCE II 400 spectrometer operating at 400 MHz (1H) in deuterated dimethyl sulfoxide. Chemical shifts are reported in ppm (d) relative to tetra methyl silane. Proton spectra were typically obtained at room temperature. Elemental analyses (C, H and N) were undertaken with a shimadzu's FLASHEA112 analyzer and all analyses were consistent with theoretical values (within  $\pm$  0.5%) unless indicated. For TLC, plates coated with silica gel were run in benzene/methanol, n-hexane/ethyl acetate mixture and spots were developed in iodine chamber. The anti-mycobacterial activities were evaluated against Mycobacterium Tuberculosis H37Rv using the tube dilution method. This methodology is nontoxic, uses a thermally- stable reagent. All the synthesized compounds were dissolved, separately, in dimethyl sulfoxide to prepare a stock solution containing 1000 µg/mL. The successive concentrations like 500, 200, 100, and 50 and µg/mL so on were prepared in a similar manner up to 6 dilutions. A sweep of Mycobacterial tuberculosis H37Rv strain culture was discharged with the help of 22 S.G.W. nichrome wire loop with a 3mm external diameter, into a sterile distilled bijou bottle containing six 3mm glass beads and 4 ml distilled water.

Antibacterial activity was assessed against *B. subtilis* (ATCC-6633) and *S. aureus* (ATCC-6538). MIC was determined using Cup-plate method. Test solution was prepared by dissolving 5 mg of the synthesized compound to 1 ml of sterile dimethyl sulphoxide (DMSO) to obtain a concentration of 5000  $\mu$ g/ml. Each Petri dish containing Muller-Hinton agar medium was inoculated with one bacterial culture by spreading the suspension of the organism with a sterile glass rod with a bended tip. In each plate cups of 6 mm diameter were made at equal distances using sterile cork borer. One cup was filled with 0.01 ml of standard drug i.e. linezolid, One was filled with 0.01 ml of DMSO; others were filled with 10  $\mu$ l-40  $\mu$ l (0.01-0.04 ml) of synthesized compound's solution in sterile DMSO. All plates were kept in the refrigerator for 30 minutes to allow the diffusion of sample to the surrounding agar medium. The Petri dishes were incubated at 37°C for 24 hrs. Diameter of the zone of inhibition was measured and the average diameter for each sample was calculated.

In the present work total 11 derivatives of N-((2-hydroxynaphthalen-1-yl)(substitutedphenyl)methyl)-4-morpholinobenzamide**3(a-k)**were synthesized.

All the compounds were identified by spectral data.

# 4.1 Synthesis of 4-morpholino benzonitrile (1)

I) Conventional Method: A mixture of morpholine (3 g, 34 mmol) and 4-chlorobenzonitrile (1.55 g, 11.2 mmol) were heated at 120  $^{0}$ C. The conversion of the 4-chlorobenzonitrile was completed in 12 hours. Then water (10 ml) was added into the reaction mixture. Then precipitate was filtered off, washed with water and dried under vacuum (30  $^{0}$ C). The compound was recrystallized with 50% aqueous ethanol.

II) Microwave-assisted Method: A mixture of morpholine (3 g, 34 mmol) and 4-chlorobenzonitrile (1.55 g, 11.2 mmol) was placed in Erlenmeyer flask and was irradiated under microwave for 25 min (high power 700 W). The completion of reaction was monitored by TLC. Then water (10 ml) was added into the reaction mixture. The precipitate obtained was filtered off, washed with water and dried under vacuum (30  $^{\circ}$ C) and recrystallized from 50% aqueous ethanol.

#### 4.2 Synthesis of 4-morpholino benzamide (2)

4- Morpholino benzonitrile (6 mmol) in ethanol (25 mL) and aqueous 6 N sodium hydroxide solution (15 ml) and was cooled to  $0^{\circ}$ C. To the above solution, 30% H<sub>2</sub>O<sub>2</sub> (20 mmol) was added and the reaction mixture was stirred at 40-50°C for 4 h. The completion of the reaction was monitored by TLC. The reaction mixture was then again cooled to  $0^{\circ}$ C, acidified with 3N H<sub>2</sub>SO<sub>4</sub> solution. After evaporation of ethanol from the reaction mixture, the residue was

extracted with  $CH_2Cl_2$ , washed with water and the organic layer was dried over  $Na_2SO_4$  and concentrated to get the desired intermediate 4-morpholino benzamide.

**4.3.1 Synthesis of N-((2-hydroxynaphthalen-1-yl) (methoxy phenyl) methyl)-4-morpholino benzamide (3 a-k)** One pot synthesis of the title compounds was achieved by stirring a mixture of 4-morpholino benzamide (1.1 mmol), 4-substitutedbenzaldehyde (1 mmol),  $\beta$ -naphthol (1 mmol) and oxalic acid (0.1 mmol) as a catalyst, on magnetic stirrer at 125<sup>o</sup>C under solvent free condition for 8-24 min. The completion of the reaction was monitored by TLC. The reaction mixture was washed with water and recrystallized from ethanol (70%). The yield and melting point of the synthesized compound was recorded.

#### Spectral data and elemental analysis data of the synthesized derivatives is given below

4.3.1 Synthesis of N-((2-hydroxynaphthalen-1-yl) (methoxy phenyl) methyl)-4-morpholino benzamide (**3** *a*) IR (KBr, v max in cm<sup>-1</sup>): 3450-3550 (O-H phenolic), 3272 (NH 2<sup>0</sup>amine), 3008 (C-C Aromatic ring), 2873 (C-H Ar alkyl), 1667(C=O amide), 1405 (C-N).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.85 (s, 3H, -OCH<sub>3</sub>), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 468, MS m/z: 468 (M). Anal. (C<sub>29</sub>H<sub>28</sub>N<sub>2</sub>O<sub>4</sub>) Calc C 74.34, H 6.02, N 5.98, Found: C 74.14, H 6.16, N 5.87.

4.3.2 Synthesis of N-((2-hydroxynaphthalen-1-yl) (hydroxy phenyl) methyl)-4-morpholino benzamide (**3** b) IR (KBr, v max in cm<sup>-1</sup>): 3416 (NH 2<sup>0</sup>amine), 3250-3350 (O-H phenolic), 3059 (C-C Aromatic ring), 2971 (C-H Ar alkyl), 1639 (C=O amide), 1410 (C-N).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 4.71 (s, 1H, -OH phenolic), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D<sub>2</sub>O exchangeable). Mol. Weight: 454, MS *m*/*z*: 455 (M+1). Anal. (C<sub>28</sub>H<sub>26</sub>N<sub>2</sub>O<sub>4</sub>) Calc C 73.99, H 5.77, N 6.16 Found: C 73.84, H 5.66, N 6.27.

4.3.3 Synthesis of N-((2-hydroxynaphthalen-1-yl)((p-tolyl) phenyl) methyl)-4-morpholino benzamide (**3** c) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH  $2^{0}$  amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 1680 (C=O amide), 1415 (C-N).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.11 (s, 3H, -CH<sub>3</sub> of Ar alkyl), 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 452, MS m/z: 453 (M+1). Anal. (C<sub>29</sub>H<sub>28</sub>N<sub>2</sub>O<sub>3</sub>) Calc C 76.97, H 6.24, N 6.19. Found: C 76.84, H 6.66, N 6.27.

4.3.4 Synthesis of N-(furan-2-yl (2-hydroxynaphthalen-1-yl) methyl)-4-morpholino benzamide (**3** d) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH 2<sup>0</sup>amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 1680 (C=O amide), 1415 (C-N).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.26 (d, 1H, -CH furan ring), 6.46 (t, 1H, -CH furan ring), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 428, MS m/z: 429 (M+1). Anal. (C<sub>26</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub>) Calc C 72.88, H 5.65, N 6.54. Found: C 72.84, H 5.76, N 6.37.

4.3.5 Synthesis of N-((2-hydroxynaphthalen-1-yl) ((2-mercaptophenyl) methyl)-4-morpholino benzamide (**3** *e*) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH 2<sup>0</sup>amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 2555 (S-H mercapto), 1680 (C=O amide), 1415 (C-N).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.40 (s, 1H, -SH mercapto) 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). MS m/z: 471 (M+1). Anal. (C<sub>28</sub>H<sub>26</sub>N<sub>2</sub>O<sub>3</sub>S) Calc C 71.46, H 5.57, N 5.95. Found: C 71.54, H 5.66, N 5.87. Mol. Weight: 470

4.3.6 Synthesis of N-((2-hydroxynaphthalen-1-yl) ((2-hydroxyphenyl) methyl)-4-morpholino benzamide (**3** *f*) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH  $2^{0}$  amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 1680 (C=O amide), 1415 (C-N).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.35 (s, 1H, -OH phenolic), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 454, MS m/z: 455 (M+1). Anal. (C<sub>28</sub>H<sub>26</sub>N<sub>2</sub>O<sub>4</sub>) Calc C 73.99, H 5.77, N 6.16. Found: C 73.84, H 5.66, N 6.27.

4.3.7 Synthesis of N-((4-chlorophenyl) (2-hydroxynaphthalen-1-yl) methyl)-4-morpholino benzamide (**3** g) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH 2<sup>0</sup>amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 1680 (C=O amide), 1415 (C-N), 760 (C-Cl).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 472, MS m/z: 473 (M+1). Anal. (C<sub>28</sub>H<sub>25</sub>ClN<sub>2</sub>O<sub>3</sub>) Calc C 71.10, H 5.33, N 5.92. Found: C 71.24, H 5.56, N 5.87.

4.3.8 Synthesis of N-((2, 4-dichlorophenyl) (2-hydroxynaphthalen-1-yl)methyl)-4-morpholino benzamide (**3** h) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH 2<sup>0</sup>amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 1680 (C=O amide), 1415 (C-N), 765 (C-Cl).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 507, MS m/z: 508 (M+1). Anal. (C<sub>28</sub>H<sub>24</sub>Cl<sub>2</sub>N<sub>2</sub>O<sub>3</sub>) Calc C 66.28, H 4.77, N 5.52. Found: C 66.34, H 4.64, N 5.67.

4.3.9 Synthesis of N-((2-hydroxynaphthalen-1-yl) (4-nitrophenyl) methyl)-4-morpholino benzamide (**3** i) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH 2<sup>0</sup>amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 1680 (C=O amide), 1415 (C-N).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 483, MS m/z: 484 (M+1). Anal. (C<sub>28</sub>H<sub>25</sub>N<sub>3</sub>O<sub>5</sub>) Calc C 69.55, H 5.21, N 8.69. Found: C 69.64, H 5.66, N 8.27.

4.3.10 Synthesis of N-((2-hydroxynaphthalen-1-yl) (phenyl) methyl)-4-morpholino benzamide (**3** *j*) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH  $2^{0}$ amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 1680 (C=O amide), 1415 (C-N).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 438, MS m/z: 439 (M+1). Anal. (C<sub>28</sub>H<sub>25</sub>N<sub>2</sub>O<sub>3</sub>) Calc C 76.69, H 5.98, N 6.39. Found: C 76.84, H 5.76, N 6.47.

4.3.11 Synthesis of N-((4-fluorophenyl)(2-hydroxynaphthalen-1-yl)methyl)-4-morpholinobenzamide (3 k) IR (KBr, v max in cm<sup>-1</sup>): 3566 (NH 2<sup>0</sup>amine), 3250-3350 (O-H phenolic), 3094 (C-C Aromatic ring), 2950 (C-H Ar alkyl), 1680 (C=O amide), 1415 (C-N), 1200 (C-F).

<sup>1</sup>H NMR (DMSOd6, 400 MHz) δ ppm: 2.93 (t, 4H, -CH<sub>2</sub> morpholine ring), 3.61 (t, 4H, -CH<sub>2</sub> morpholine ring), 5.51 (s, 1H, -OH β-naphthol), 5.82 (s, 1H, -CH Ar alkyl), 6.80-8.20 (m, 14H, Ar-H), 9.05 (s, 1H, -CONH, D2O exchangeable). Mol. Weight: 456, MS m/z: 457 (M+1). Anal. (C<sub>28</sub>H<sub>25</sub>FN<sub>2</sub>O<sub>3</sub>) Calc C 73.67, H 5.52, N 6.14. Found: C 73.84, H 5.66, N 6.27.

#### Pharmacology

#### **Biological investigation**

All the synthesized compounds were assayed in-vitro for antibacterial activity against *E. coli* (ATCC 8739), *P. aeruginosa* (ATCC 9027), *S. aureus* (ATCC 6538), *B. subtilis* (ATCC 6633) using Linezolid as a reference standard. Test solution was prepared by dissolving 5 mg of the synthesized compound to 1 ml of sterile dimethyl sulphoxide (DMSO) to obtain a concentration of 5000  $\mu$ g/ml. The compounds were screened for anti tubercular activity against *Mycobacterium tuberculosis* H<sub>37</sub>RV strain using Isoniazid, Rifampicine as a reference standard. Test solution was prepared by dissolving 10 mg of the synthesized compound to 10 ml of sterile dimethyl sulphoxide (DMSO) to obtain a concentration of 1000  $\mu$ g/ml [18-22].

#### **RESULTS AND DISCUSSION**

Total 11 derivatives of N-((2-hydroxynaphthalen-1-yl) (substituted phenyl)methyl)-4-morpholino benzamide were synthesized and screened for their antibacterial activity using Tube Dilution Technique (Linezolid as a standard). Some of the synthesized compounds like **3b** (R= 4-Hydroxy) have shown good activity against *S. aureus* and *B. subtilis*, **3g** (R= 4-Chloro) and **3k** (R= 4-Fluoro) have shown excellent activity against *S. aureus* and *B. subtilis* when compared with standard Linezolid. MIC values are given in **Table 3.** 

Compound	Code	S. aureus ATCC	B .subtilis ATCC
		6538	6633
3a		>200	>200
3b		<b>≤150</b>	≤ <b>150</b>
3c		>200	>200
3d		>200	>200
3e		>200	>200
3f		>200	>200
3g		<b>≤ 100</b>	≤ <b>100</b>
3h		>200	>200
3i		>200	>200
Зј		>200	>200
3k		<b>≤ 50</b>	<b>≤ 50</b>
Std (Linezolid)		<b>≤ 30</b>	<b>≤ 30</b>

#### Table 3: Data of Antibacterial activity (MIC µg/ml)

The result of antitubercular activity shows that compound 3h (R= 2, 4-Dichloro) has shown excellent antitubercular activity. Compound 3f (R= 2-hydroxy) has shown significant antitubercular activity. Compound 3i (R= 4-Nitro) has shown comparable antitubercular activity. All other compounds have shown satisfactory antitubercular activity when compared with the results obtained from standard. The result of antitubercular evaluation is given in **Table 4**.

Table 4: Data of Antitubercular activity (MIC µg/ml)

Compound Code	MIC µg/ml
3a	1000
3b	500
3c	250
3d	1000
3e	250
3f	62.5
3g	250
3h	50
3i	100
3j	250
Isoniazid	0.2
Rifampicin	40

### CONCLUSION

Total 11 derivatives of N-((2-hydroxynaphthalen-1-yl) (substitutedphenyl) methyl)-4-morpholinobenzamide **3(a-k)** were synthesized as per the scheme reported. Structures of synthesized compounds were confirmed by spectral study such as IR, <sup>1</sup>HNMR and Mass, Elemental analysis. The synthesized compounds were subjected to antibacterial and antitubercular evaluation. Some of the synthesized compounds like 3b (R= 4-Hydroxy) has shown comparable activity against *S. aureus* and *B. subtilis*, 3g (R= 4-Chloro) and 3k (R= 4-Fluoro) have shown excellent activity against *S. aureus* and *B. subtilis*, 3g (R= 4-Chloro) and 3k (R= 4-Fluoro) have shown excellent activity shows that compound **3h** (R= 2, 4-Dichloro) has shown excellent antitubercular activity. Compound **3f**(R= 2-hydroxy) has shown significant antitubercular activity. Compound **3i** (R= 4-Nitro) have shown comparable antitubercular activity. All other compounds have shown satisfactory antitubercular activity when compared with the results obtained from standard (Rifampicin).

The correlation of activity with the structure of synthesized derivatives, it has been observed that electron donating groups like 4-Chloro (**3g**), 2, 4-Dichloro (**3h**), 4-Fluoro (**3k**), attached to the phenyl ring increases antimicrobial activity. When the substituent Ar has electron donating groups such as 4-Hydroxy (**3b**), 2-Hydroxy (**3f**) showed moderate antimicrobial activity. The structure-activity relationship study showed that the morpholine ring is essential for antibacterial activity. The antibacterial activity may increase or decrease depending upon substituent Ar

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groups on aromatic ring. The structure -activity relationship revealed amide linkage (-CONH) to be essential for antitubercular activity.

The presence of two important moieties i.e. Morpholine and  $\beta$ -naphthol in the final derivatives containing –CONHgroup have contributed towards better antimicrobial activity. Thus the designed N-((2-hydroxynaphthalen-1yl)(substitutedphenyl)methyl)-4-morpholino benzamide derivatives have shown significant antimicrobial activity.

From the research work undertaken and results obtained, it appears that, N-((2-hydroxynaphthalen-1-yl) (substituted phenyl) methyl)-4-morpholinobenzamide derivatives possess very good potential for antimicrobial activity and they can be developed as potent chemotherapeutic agents.

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#### REFERENCES

- [1] Prescott L, Harley J, Klein D, Microbiology, The McGraw Hill companies, 2002, pp 16.
- [2] Dalhoff A, Infection, 1994, 22, 111.
- [3] Livermore D, Int. J. Antimicrob. Agents, 2000, 16, S3.
- [4] Lee V, Hecker S, J. Med. Chem, 1999, 19, 521.

[5] Koca M, Servi S, Kirilimis C, Ahmedzade M, Kazaz C, Ozbek B, Otuk G, Eur. J. Med. Chem. 2005, 40, 13.

[6] Sharma R, Sharma K, Dikshit S, Advances in Applied Science Research, 2011, 2 (1), 178-192.

- [7] Saleshier F, Suresh S, Anitha N, Karim J, European Journal of Experimental Biology, 2011, 1 (2), 150-159.
- [8] Sharma R, Sharma K, Dixit S, Der Chemica Sinica, 2010, 1 (1), 57-66.

[9] Wilson and Gisvolds, *Textbook of organic medicinal and pharamaceutical chemistry*, Lippincott, Williams and Wilkins, **2004**, pp 255-257.

[10] Mkpenine V, Ebong G, Oboy I, Abasiekong, 2008, 5,431.

- [11] Faizul A, Satendra S, Sukhbir Lal K, Zhejiang J Univ Sci. B, 2007, 8,446.
- [12] Chung K, Chase M, Arch Pharm Res, 2005, 28, 750.
- [13] Huang M, Pyang W, Drug Development Research, 2003, 60,261.
- [14] Perumal P, Rajasree R, Eur. J. Med. Chem, 2005, 40, 225–229.
- [15] Raparti V, Chitre T, Bothra K, Kumar V, Eur. J. Med. Chem, 2009, 44, 3954–3960.
- [16] Lakshminarayana N, Rajendra Prasad Y, Eur. J. Med. Chem, 2010, 45, 3709-3718.
- [17] Shinde D, Nagawade R, Mendeleev Commun., 2007, 17 299-300.
- [18] Nikalje A, Pathan M, Narute A, Ghodke M, Rajani D, Der Pharmacia Sinica, 2012, 3 (2) 229-238
- [19] Indulatha V, Gopal N, Jayakar B, Der Chemica Sinica, 2011, 2(6),48-57.
- [20] Subramaniam R, Rao G, Pai S, Nagesh P, Der Pharmacia Sinica, 2011, 2 (3), 146-155.
- [21] Solankee A, Patel G, Patel R, Patel K, Der Chemica Sinica, 2010, 1 (2), 85-91.
- [22] 22. Valarmathi R, Akilandeswari S, Indu latha V, Umadevi G, Der Pharmacia Sinica, 2011, 2 (5) 64-68.