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Coherent-synchronized oxidation of methane by hydrogen peroxide to methanol on a biomimetic catalyst

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Abstract

The paper is devoted to the synthesis and study of the structure, physicochemical properties and activity of a biomimetic catalyst, which is a model of the heme iron-containing enzyme cyto-chrome P-450, in the reaction of direct conversion of methane to methanol by hydrogen peroxide. The biomimetic catalyst consists of iron pentafluorotetraphenylporphyrin applied to a mesoporous solid carrier of acid-base nature Al2O3. A study of the specific surface area, adsorption capacity, structure, character and sizes of the carrier pores, which play an important role in the synthesis and activity of the biomimetic catalyst, and measured as functions of relative or absolute pressure using a fully automated 3Flex analyzer, was conducted. The adsorption was carried out with liquid nitro-gen at a temperature of 77.3 K. As a result, the value of the specific surface area of the Al2O3 carrier measured by BET method was 239 m2/g, while the catalyst was 216.9 m2/g. The mesoporosity of the carrier and its pore sizes were determined using the BJH method; pores with an average diameter of 2.42-22.2 μ m have maximum total volume of V = 0.35-0.33 cm3/g. Mesopores with a diameter of 11.26-22.2 μ m have a total volume of 0.132-0.0135 cm3/g and correspond to the size and structure of the iron porphyrin complex (16-18 μ m).

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Biography

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