

## Solvent Based Variations in Yield of Bioactive Extracts from the Sclerotium of Pleurotustuber-regium

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### ABSTRACT

**Background:** The therapeutic effectiveness of herbs and fungi used for medicinal purposes is not only a factor of their bioactive constituents but also a factor of both the extraction solvent and extraction method.

**Objective:** The objective of this study is to extract and analyze the bioactive components in the sclerotium of Pleurotus tuber-regium using different solvents, as to ascertain the solvent that gives a better yield.

**Method:** A quantity of 10.0 kg of fresh sclerotia of P. tuber-regium purchased at Zarama Market in Southern Nigeria was washed, peeled and the white inner parts were sliced using a sterilized knife. The sliced samples were dried at room temperature for fourteen days. After grinding, the bioactive components were extracted by weighing 10 g (RI VDP SOH LQWR WKUHH ZHOO VWRSSHU ERWVOHV DQG HDFK ZDV H hexane and dichloromethane), while that of soxhlet extraction was done in a soxhlet apparatus, using ethanol as the VROYHQW 7KH SURFHVV ZDV UHSHDWHG WZLFH DQG WKH FRPELQHG D Two milliliters of the extracts were used for gas chromatographic and mass spectroscopy analysis.

**Result:** The highest peak on the chromatogram for the methanol extract was observed at 32.644 min., while hexane, dichloromethane and soxhlet extracts had their highest peaks at 31.459 min., 14.254 min. and 18.060 min. respectively. 7KH KLJKHVW ELRDFWLYH FRPSRQH QW LQ PHWKDQRO H[WUDFW ZDV + D PHWKDQH ZLWK D YDOXH RI ZKLOH KH[DQH GLFKORUR WHWUDGHFDPHWK\O %LV HWK\OKH\O SKWKDODWH DQG 3KWKDOL DQG UHVSFWLYHO\

**Conclusion:** Hexane and dichloromethane extracts yielded more bioactive components with better nutraceutical and medicinal properties and may be regarded as better solvents for mushroom and fungi extractions..

**Keywords:** Mushroom, Bioactive components, Extraction methods, Solvents, extracts

### INTRODUCTION

Different plants and fungal materials have been used in traditional medicine to solve varieties of health problems. The effectiveness of various combinations of herbs and other ingredients used by traditional medical practitioners depends on the constituents of the selected plant or fungi materials. The availability of the constituents is a factor of the adopted extraction method. The extraction of a given compound from a substrate depends on both the solubility of the compound and the polarity of the solvent. Modern extraction techniques such as accelerated solvent extraction, VXSHUFULWLFDO AXLG H[WUDFWLRQ PLFURZDYH DVVLVWHG H[WUDFW in nutraceuticals extractions from plants. These modern techniques are mainly targeted at decreasing extraction time, ZKLOH LQFUHDVLRQ \LHOG DQG HQKDQFLQJ H[WUDFW TXDOLW\ ZLW WRWDO SKHQROLF FRQWHQW WRWDO ADYRQ Murriga-Fste-Opeta-@ves DQG DQV RSWLPL]LQJ XOWUDVRQLF DVVLVWHG H[WUDFWLRQ XVLQJ UHVSQRVH PLQ HWKDQRO FRQFHQWUDWLRQ DQG VROYHQW WR VDP SOH UDW

2LO H[WUDFWLRQ LV PDLQO\ D GLIIXVLRQ SURFHVV LQ ZKLFK WKH V sample material, thereby forming a solution of the oil in the solvent. Though some of these plants and fungal materials DUH VWLOO YHU\ UHOHYDQW LQ WUDGLWLRQDO PHGLFLQH IRU WUHDV RI WKHLU LVRODWHV KDYH VKRZQ SRWHQWLDO DQWL LQADPPDWRU\ D

Pleurotus tuberregium is a notable saprotroph that produces a food storage sclerotium upon its consumption of GHFD\LQJ ZRRG > @ ,W LV RQH PXVKURRP WKDW ERWK LWV EDVLGLF WKHLU QXWULWLYH DQG PHGLFLQDO SURSHUWLHV > @ 7KH H[WUDFW IRU WKH WUHDWPHQW DQG PDQDJHPHQW RI GLVHDVHV UDQJLQJ IURF KHDGDFKH VWRPDFK SUREOHP FROG DVWKPDIHYHU KLJK EORRG /H%DXHU > @ UHSRUWHG WKH XVH RI WKLV PXVKURRP DQG LWV H[WU VRXS DQG IRU WKH WUHDWPHQW RI KHDUW UHODWHG DLOPHQWV Z FRXJK DQG REHVLW\ LQ WKH 6RXWKHUQ SDUWbe Regium DOWLD JU@Z\$V LQG\$ DQG \$XVWUDOLD > @

2WKHU WKHUDSHXWLF DFWLYLWLHV VXFK DV DQWLWXPRXU LP hypocholesterolaemic, antihypertensive, antihyperglycaemic, antimicrobial and antiviral acPleurotus spp. KDV DOVR EHHQ UHSRUWHG > @ 7KRXJK WKHVH WKHUDSHXWLF DFWL fermentation broth, mycelia and fruiting bodiesPleurotus VSS > @ WKH ELRFKHPLFDO PHFKD GXH SRRU FKDUDFWHUL]DWLRQ DQG LGHQWL¿FDWLRQ RI WKH ELRDFW bioactive components in the sclerotiumPleurotus using different solvents, to analyze the extracts obtained for LGHQWL¿FDWLRQ RI WKH ELRDFWLYH FRPSRQHQWV DQG WR DVFHUW

## MATERIALS AND METHODS

### Sample collection, preparation and extraction

\$ TXDQWLW\ RI NJ Pleurotus tuberregium SXURFWDDHG DW =DUDPD ODUNHW LQ ZDVKHG SHHOG DQG WKH ZKLWH LQQHU SDUWV ZHUH VOLFHG XVLQ DW URRP WHPSHUDWXUH LQ D GXVW IUHH HQYLURQPHQW IRU D SHUL EOHQGHU LQWR D ¿QH SRZGHU 8VLQJ DQ DQDO\WLFDO ZHLJKLQJ EDZHO VWRSSHU ERWROHV \$ YROXP RI P/ RI D VSHFL¿F H[WUDFWL DGGHG WR HDFK VWRSSHU ERWROH 7KH PL[WXUHV ZHUH YLJRURXVC VR[OHWK H[WUDFWLRQ ZDV GRQH LQ D VR[OHWK DSSDUDWXV XVLQJ ZDV FROOHFWHG E\ ¿OWHULQJ LQWR D TXDUW] EHDNHU LQ D IXPH K DOLTXRW FROOHFWHG IURP HDFK H[WUDFWLRQ VROYHQW ZHUH VHSD E\ SDVVLQJ WKURXJK D SDVWXUH SLSHWWH SDFNHG ZLWK VLOLFDD JHO DLU GULHG WR P/ IRU JDV FKURPDWRJUDSKLF DQDO\LV

### GC-MS analysis of extracts

7KH H[WUDFWV ZHUH DQDO\HG XVLQJ D FRPELQHG JDV FKURPDWRJUD \$JLOHQW 7HFK ¿WWHG ZLWK D FDSLOODU\ FROXPQ +3 06 SKH XVLQJ +HOLXP DV D FDUULHU JDV DW LQLWLDO FROXPQ WHPSHUDWXU ZDV LQFUHDVHG DW f& SHU PLQXWHV WR f& DQG KHOG IRU PLQX ZDV GRQH DW LRQL]DWLRQ HQHUJ\ RI H9 7KH RLO ZDV GLOXWHG DXWRPDWLFDOO\ LQMHFWHG LQWR \$JLOHQW 7HFK PRGHO PDVV V XVLQJ WKH &HP 2I¿FH VRIWZDUH DWDFKHG WRWKH 06 OLEUDU\ 7 FRQ¿UPHG XVLQJ WKH GDWDEDVH R7HFKVIRRODU ,Q¿WLWXWH RI 6WDQ

## RESULTS

7KH &KURPDWRJUDP RI ELRDFWLYH FRPSRQHQWV RI PHWKDQRO KH sclerotia of *P. tuberregium* DUH VKRZQ LQ UHVSHFWLYHO\ 7KH KLJKHVW SHDN IR REVHUYH DW PLQ +H[DQH H[WUDFW KDV LWV KLJKHVW SHDN DW KDYH WKHLU KLJKHVW SHDNV DW PLQ DQG PLQ UHVSHFWL

7KH UHWHQWLRQ WLP SHUFHQW DJH FRQFHQWUDWLRQ PROHFODU components of methanol, hexane, dichloromethane and soxhlet extracts of the sclerotium *P. tuberregium* DUH VKRZQ in Tables 1-4 D5 5 5 7HWUDPHWK\O KH[DK\GUR + D FRPSRQHQW LQ WKH PHWKDQRO H[WUDFW ZLWK D YDOXH RI IR

Figure 1: &KURPDWRJUDP RI ELRDFWLYH FRPSRQHONW RPHWKDQRO H[WUDF

Figure 2: &KURPDWRJUDP RI ELRDFWLYH FRPSRQHONW RPHWKDQRO H[WUDF

Figure 3: &KURPDWRJUDP RI ELRDFWLYH FRPSRQHQQWV R. ~~Table 3~~ GLEFORURPDWKDQ

Figure 4: &KURPDWRJUDP RI ELRDFWLYH FRPSRQHQQWV R. ~~Table 4~~ VVRK[OHW H[WUDFW

HVWHU EHW D :LWK D YDOXH RI +H[ DQH H[WUDFW KDV L  
 WHWUDGHFDPHWK\O ZLWK D YDOXH RI IROORZHG E\ +H[DFRVI  
 %LV HWK\OKH[\O SKWKDODWH DQG (LFRVDQH ZHUH WKH KLJKHVW E  
 ZLWK YDOXH RI DQG UHVSHFWLYHO\ 7KH KLJKHVW ELRDFV  
 3KWKDOLF DFLG FKORUREHQ]\O EXW\O HVWHU ZLWK D YDOXH RI  
 GLPHWKR[\ ZLWK D YDOXH RI

DISCUSSION

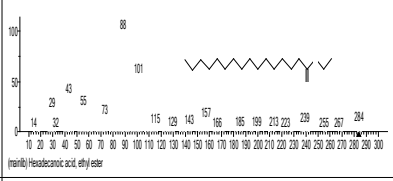
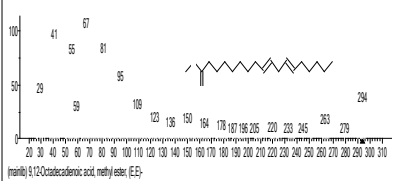
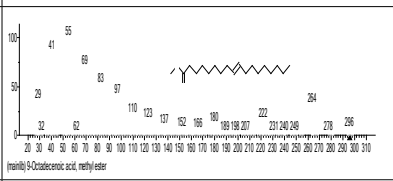
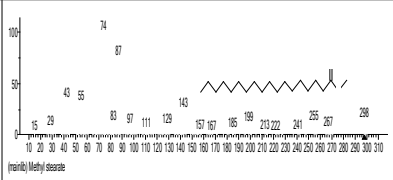
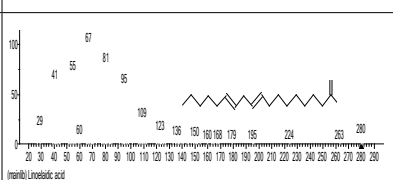
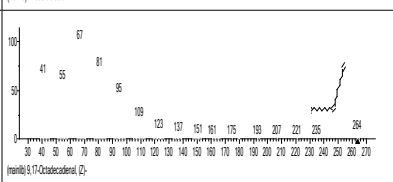
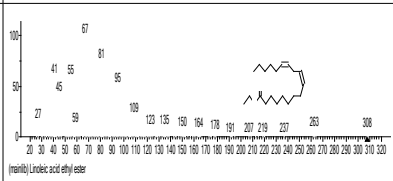
The disparity in both compound quantity and concentration observed in the methanol, hexane, dichloromethane and soxhlet extracts of the sclerotia of tuberregium VKRZV YDU\LQJ FDSDELWLHV RI GLIIHU  
 liberate different compounds from a substrate. The ability of methanol to extract only three uncommon polycyclic  
 FRPSRXQGV + &\FORSURSD>D@QDSKWKDOHQ RQH D5 5 5  
 PHWKDQH DQG 8UV HQ RLF DFLG K\GUR[\ PHWK\O HVWHU  
 1 indicates that the use of methanol as an extraction solvent may favour the extraction of polycyclic compounds  
 HVSHFLDOO\ LQ D OLJQLQ EDVHG VXEVDWH OLNH PXVKURRP \$V  
 H[WUDFWV LQ PHWKDQRO H[WUDFWLRQ PD\ EH ORZHU WKDQ LQ D QRO  
 ODUJHU SURSRUWLRQ RI SRODU FRPSRXQGV +RZHYHU LWV VROXELO  
 polar compounds. The result of this study is an indication that methanol may not be a solvent of choice for extraction  
 ZKHUH ERWK SRODU DQG QRQ SRODU FRPSRXQGV DUH QHHGHG IURP  
 \$OWKRJK GLIIHUHQW VROYHQWV KDV EHHQ XVHG IRU H[WUDFWLRQ  
 EHFDXVH RI LWV QRQ SRODU SURSHUWLHV ORZ ERLQLQ SRLQW f  
 &ORJK DQG 0XOKROODQG > @ DOVR UHSRUWHG WKDW WKH ORZ UHDF  
 IRU WKH H[WUDFWLRQ RI UHDFWLYH FRPSRXQGV 7KH KH[DQH H[WUDF  
 GLIIHUHQW ELRDFWLYH FRPSRQHQWV PDGH XS RI OL (Table 2) The E UDQFKI  
 ODUJH QXPEHU RI FRPSRXQGV REVHUYHG LQ WKH KH[DQH H[WUDFW  
 release these compounds from the substrate.

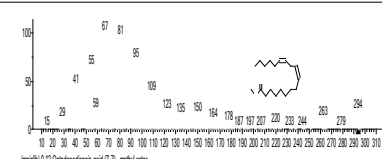
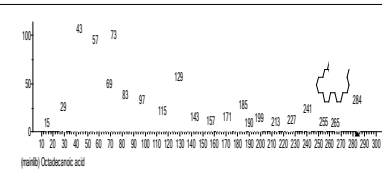
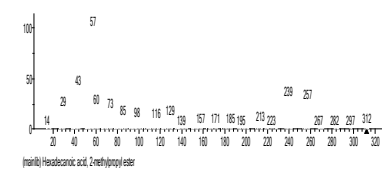
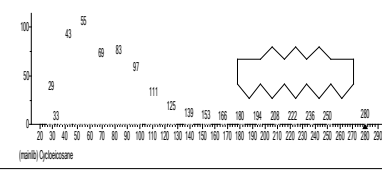

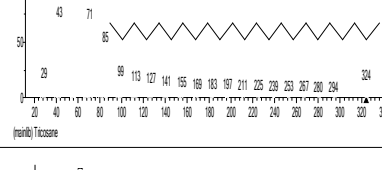
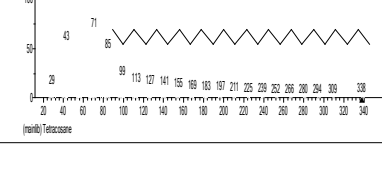
7KH FRPSRQHQW REVHUYHG IURP Table 1, KRLZFK OVRKURAP HWWKHID Q RI OH[WLLQ  
 GLFKORURPHWKDQH WR GLVVROYH D ZLGH UDQJH RI RUJDQLF FRPSRX  
 > @ 7KH SUHVHQFH RI ERWK OLQHDU DQG F\FOLF FRPSRXQGV LQ WK  
 potentials to penetrate and extract most constituent compounds from a lignin based sample. Though dichloromethane  
 LV WKH OHDVW WR[LFDPRQJWKH VLPSON FKORURK\GURFDUERQV  
 H[WUDFWLRQ > @ \$V DQ H[WUDFWLRQ SURFHVV VR[KOHV H[WUDFW

S/N	Compound	Retention Time (min)	Percentage of the total	Molecular formula	Molecular weight	Structure
	+ &\FORSURSD>D@QDSKWKDOHQ RQH			& +		
	D5 5 5 7HWUDPHWK\O KH[DK\GUR + D methane			& + 1 <sub>3</sub>		
3	8UV HQ RLF DFLG K\GUR[\ PHWK\O HVWHU EHW D			& + O <sub>3</sub>		

Table 1: Bioactive components of methanol extract of sclerotia of tuberregium.

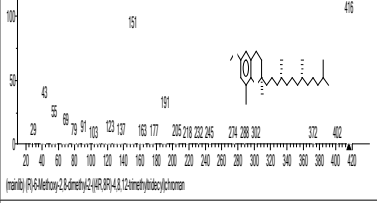
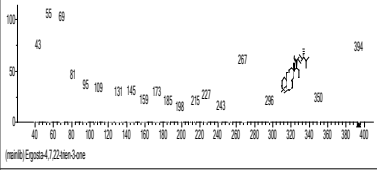
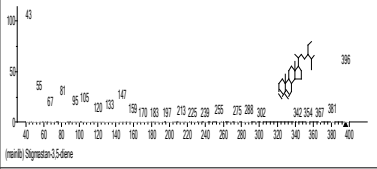
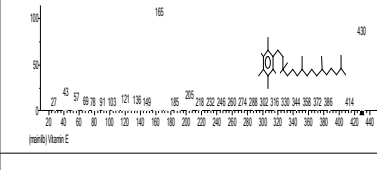
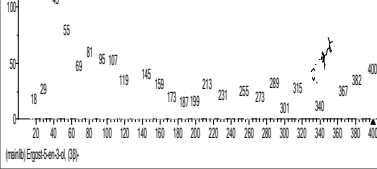
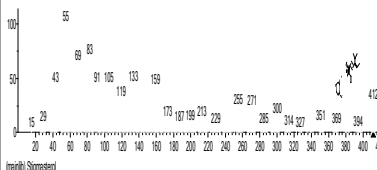
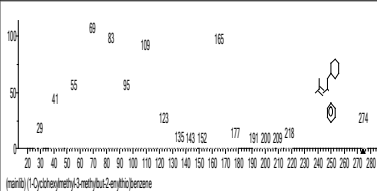
S/N	Compound	Retention Time (min)	Percentage of the total	Molecular formula	Molecular weight	Structure
	& \FORGRGHFDQH			& +		
	1RQDQH PHWK\O			& +		
3	8QG HFDQH			& +		
	'RGHFDQH	WULPHWK\O		& +		
	2FWDG HFDQH	FKORUR		& + <sub>37</sub> & O		
6	Tetratetracontane			& +		
7	+HSWDPHWK\O ELV WULPHW K\OVLOR[\ WHWUDVLOR[ DQH			& + O Si <sub>6</sub>		
8	+H[DG HFDQRLF ester	DFLG PHWK\O		& + O		
	Q +H[DG HFDQRLF DFLG			& + O		

	&\FORKHSWDV LOR[DQH tetradeca PHWK\O			& + O <sub>7</sub> Si <sub>7</sub>	
	+H[DGHF DQRLF DFLG ester HWK\O			& + O <sub>36</sub>	
	2FWDG HFDGLHQRLF acid,methyl HVWHU ( (			& + O	
	2FWDG H FHQRLF DFLG HVWHU (			PHWK\O & + O <sub>36</sub>	
	Methyl stearate			& + O <sub>38</sub>	
	+HSWDV LOR[DQH KH[DG HFD PHWK\O			& + O <sub>6</sub> Si <sub>7</sub>	
	/LQRHODLGLF DFLG			& + O	
	2FWDG HFDGLHQDO =			& + O	
	/LQROHLF DFLG HWK\O HVWHU			& + O <sub>36</sub>	

	2FWDGHFDGLHQRLF DFLG = =		3.668	& + O	
	Octadecanoic acid			& + O <sub>36</sub>	
	+H[DGHFQRLF DFLG PHWK\O propyl ester			& + O	
	&\FORHLFRVDQH			& +	
	&\FORQRQDVLOR[ PHWK\O	DQH RFWDGHFD		& + O <sub>Si</sub>	
	Tricosane			& +	
	Tetracosane			& +	
	,VRSURSR[\ +HSWDPHWK\O K\OVLOR[\ WHWUDVLOR[ DQH	ELV WULPHW		& + O <sub>7</sub> Si <sub>7</sub>	
	3HQWDFRVDQH			& +	



	'LK\GUR[\ benzoic acid, 3TMS derivative			& + O Si <sub>3</sub>	
	+H[DFRVDQH			& +	
	+HSWDFRVDQH			& +	
	&\FORQRQDVLOR[ PHWK\O	RFWDGHFD		& + O Si	
	'RFRVDQH KH\O			& +	
33	7ULPHWK\OWULGHFDQH			& +	
	&\FORQRQDVLOR[ PHWK\O	RFWDGHFD		& + O Si	
	+H[DFRVDQH	GRGHF\O		& + <sub>38</sub> <sub>78</sub>	
36	(UJRVW HQ EHWD 5	RO DFHWDWH		& + O	

37	+H[DVLOR[DQH WHWUDGHFDPHWK\O			& + OSi <sub>6</sub>	
38	5 0HWKR[\ GLPHWK\O 5 5 WULPHWK\OWUL GHF\O FKURPDQ			& + O	
	(UJRVWD WULHQ RQH			& + O	
	6WLJPDVWDQ GLHQH			& +	
	9LWDPLQ (			& + O	
	(UJRVW HQ RO EHWd			& + O	
	Stigmasterol			& + O	
	&\FORKH[\O PHWK\O PHWK\OEXW HQ\OWKLR benzene			& +	

	JDPPD 6LWRVWHURO			& + O	
	+ 1DSKWKDOHQQRQH D KH[D K\GUR D GLPHWK\O PHWK\O HWKHQ\O			& + O	
	7KLHQ\O isoxazo lemethanol	GLK\GUR		& + 1 2S	
	D5 5 5 PHWK\O KH[DK\GUR + D PHWK DQRD]XOHQ RQH	7HWUD		& + 1 <sub>3</sub>	
	7HUW %XW\O GLPHWK\OVLO\O	GLPHWK\O		& + O Si	

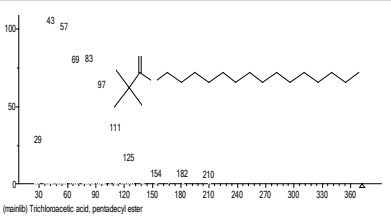
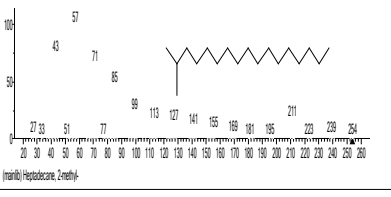
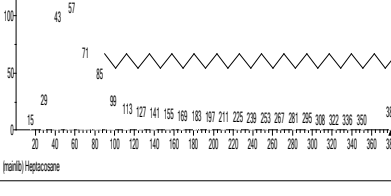
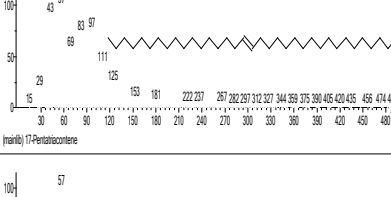
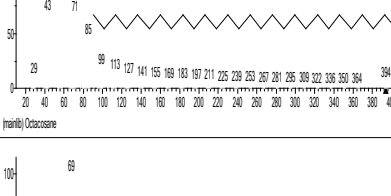
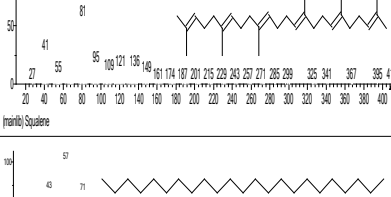
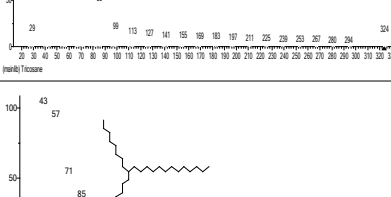
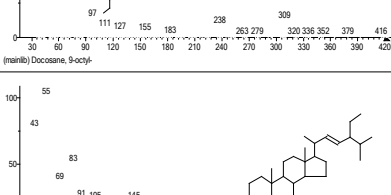
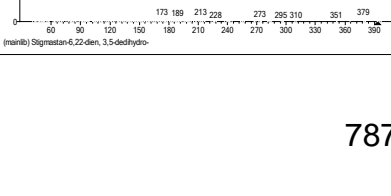
Table 2: Bioactive components of hexane extract of sclerotia of *Aspergillus terreus*

S/N	Compound	Retention Time (min)	Percentage of the total	Molecular formula	Molecular weight	Structure
	&DUERQLF KH[DGHF\O HQ \O ester	DFLG SURS		& + O <sub>3</sub>		

	&DUERQLF nonyl vinyl ester	DFL G		& + O <sub>3</sub>		<p>Mass spectrum of Nonyl vinyl ester. Base peak at m/z 43. Other significant peaks at 57, 71, 85, 99, 113, 127, 141, 155, 169, 183, 197, 211, 225.</p>
3.	\$FHWR[\ \ PHWK\ O furaldehyde			& + O		<p>Mass spectrum of 5-Acetylmethyl-2-furaldehyde. Base peak at m/z 126. Other significant peaks at 43, 57, 71, 85, 99, 113, 127, 141, 155, 169, 183, 197, 211, 225.</p>
	+ H[DG HFDQH			& +		<p>Mass spectrum of Heptadecane. Base peak at m/z 43. Other significant peaks at 57, 71, 85, 99, 113, 127, 141, 155, 169, 183, 197, 211, 225.</p>
	> @ 7HWUD]ROR> E @ > @WULD]LQH WHWUDK\GUR			& + 1 <sub>6</sub>		<p>Mass spectrum of 1,2,4,7-tetrahydro-5H-1,4-diazepine, 5,6,7-trimethyl-. Base peak at m/z 42. Other significant peaks at 56, 70, 84, 98, 112, 126, 140, 154, 168, 182, 196, 210, 224.</p>
6.	1 RQDGHFHQH			& + <sub>38</sub>		<p>Mass spectrum of Nonadecane. Base peak at m/z 43. Other significant peaks at 57, 71, 85, 99, 113, 127, 141, 155, 169, 183, 197, 211, 225.</p>
7.	Octadecane			& + <sub>38</sub>		<p>Mass spectrum of Octadecane. Base peak at m/z 43. Other significant peaks at 57, 71, 85, 99, 113, 127, 141, 155, 169, 183, 197, 211, 225.</p>
8.	%LF\FOR> HQH	@QRQ		& +		<p>Mass spectrum of Bicyclo[5.1.0]non-1-ene. Base peak at m/z 79. Other significant peaks at 53, 67, 81, 95, 109, 123.</p>
	0HWK\OHQH vinylcyclo pentane			& +		<p>Mass spectrum of 1-Methylec-2-vinylcyclopentane. Base peak at m/z 93. Other significant peaks at 39, 53, 67, 81, 109.</p>

	1 R Q D G H F D Q H			& +	
	+H[D G H F D Q R L F D F L G methyl ester			& + O	
	Dibutyl phthalate			& + O	
	Q +H[D G H F D Q R L F acid			& + O	
	'R F R V H Q H			& +	
	Eicosane			& +	
	2 F W D G H F D G L H Q R L F D F L G = = methyl ester			& + O	
	+H Q H L F R V D Q H			& +	

	2 FWDGHFD GLHQRLF DFLG = =			& + O		
	'RFRVHQH			& +		
	Docosane			& +		
	Methoxyacetic DFLG WULGHF\O est er			& + O <sub>3</sub>		
	+HSWDGHF\O WULÀXRURDFHWDWH			& + F <sub>3</sub> O		
	Tricosane, PHWK\O			& +		
	3HQWDFRVDQH			& +		
	%LV HWK\OKH[\O phthalate			& + O <sub>38</sub>		

	Trichloroacetic acid, pentadecyl ester			& + $\text{O}_3$	373.786	
	+ H S W D G H F D Q H P H W K \ O		3.866	& + <sub>38</sub>		
	+ H S W D F R V D Q H			& +		
	3 H Q W D W U L D contene			& +		
	Octacosane			& +		
	Squalene			& +		
	Tricosane			& +		
33	' R F R V D Q H	R F W \ O		& +		
	6 W L J P D V W D Q G L H Q G L K \ G U R	G H		& + O		

	2 F W D G H F D Q H sulphonyl chloride			& + <sub>37</sub> & O <sub>2</sub>	
36	1 R U F K R O H V W H Q E H W D R O R Q H			& + O	
37	Tetracontane, W U L P H W K \ O			& + <sub>88</sub>	
38	( U J R V W H Q R O E H W D			& + O	
	Stigmasterol			& + O	
	Octacosane			& +	
	E H W D 6 L W R V W H U R O			& + O	
	+ 7 U L D ] R O H F D U E R [ \ O L F D F L G D P L Q R S K H Q \ O P H W K \ O hydrazide			C <sub>11</sub> H <sub>11</sub> N <sub>3</sub> O <sub>2</sub>	
	Octacosane			& +	



<p>+ &amp; \ FORSURSD &gt; D @                  QDSKWKDOHQ                  RQH D                    D E R F W D                    K \ G U R D                  W H W U D P H W K \ O                  alpha.,7.                  alpha.,7a.alpha.,7b.                  D O S K D</p>		<p>D</p>	<p>&amp; +</p>	
<p>8 U V H Q                  D F H W D W H</p>	<p>R O                  E H W D</p>		<p>&amp; + O</p>	

Table 3: Bioactive components of dichloromethane extract of sclerotia of *P. tuberregium*

S/N	Compound	Retention Time (min)	Percentage of the total	Molecular formula	Molecular weight	Structure
1	Tricyclo[2.2.1.0(2,6)]heptan-3-one, oxime			& +		
2	Undec-10-ynoic acid, but-3-yn-2-yl ester			& + O		
3	2,5-Cyclohexadiene-1,4-dione, 2,5-dihydroxy-3-methyl-6-(1-methylethyl)-			& + O		
4	3-Cyclopropenoic acid, 1-butyl, methyl ester			& + O		
5	4-Pyridinol			& + 1 2		

6	Phthalic acid, 3-chlorobenzyl butyl ester				
7	Hexadecanoic acid, methyl ester			& + O	
8	2-Methyl-3-(phenylsulfonyl) methyl-2-cyclopenten-1-one			& + SO	
9	1,4-Naphthalenedione, 5,8-dihydroxy-2,7-dimethoxy-			& + O	
10	2-Cyclopenten-1-one, 3-methyl-			& + O	
11	n-Hexadecanoic acid			& + O	

Table 4: Bioactive components of soxhlet extract of sclerotium of *Boerhaavia diffusa*.

HLWKHU LQVROXEOH RU VSDULQJO\ VROXEOH LQ ZDWHU 7KRXJK VR[UHSHDWHG DSSOLFDWLRQ RI KRW VROYHQW HWKDQRO LQ WKLV FDVH RI WKH VROYHQW LQWR WKH VDPSON +RZHYHU WKH \LHOG RI WKH H[PD\ EH UHVSQRVLEOH IRU WKH ORZ \LHOG FRPSRXQGV REVHUYHG

CONCLUSION

+H[DQH DQG GLFKORURPHWKDQH H[WUDFWV \LHOGHG PRUH ELRDFWLYH FRPSPHWKDQRO DQG VR[KOHW H[WUDFWV 6HTXHO WR WKHVH QGLQJV KH[DQH D based substrate such as mushrooms and other fungi.

REFERENCES

'DGL ': (PLUH 6\$ +DJR 2018) P(SUR Y% PHQW RI \LHOG RI ELRDFWLYH FDFWLYLW\ RI PRULQJD VWHQRSHWDOD OHDYHV E\J R&S&T&H&J XOW

%RUFKHUV \$7 .ULVKQDPXWK\ \$ .HH(2008) The Ontology of mushrooms LQ 0 ([S %LRO OHG 0D\ZRRG

3. /XOO & :LFKHUV +- (2005) SHOWIRLOA P PDWRU\ DQG LPPXQRPRGXODW metabolites.0HGLDWRUV ,QADPP

2KLUL 5& \$PDGL (2018) 3KRMIFK HPLFDO FRQ QPHHQW LQ VHFWRQV WXEHU PXVKUR R&S&T&H&J RFDUS

&KHQ \$: +X(2004) 31URGXFWRQ RI WXEHU OLNH P&S&T&H&J RFDUS

- 
- 6 LQJ \$JDULFRIRUJFHEVMUChrbDnt\$  
6. ,VLNKXHPKHQ 62(2004)% DREUQ\$ 3OHXUR WushwvWXEh HUUHJLXP  
7. 2VR (%97) 3OHXURWXV WXEHumycollogiaXP IURP 1LJHULD  
8. \*UHJRUL \$ 6YDJH(2007) & XIKVLYDHWL RQ WHFKQLTXHV DQG Fbb6 LFLQDQ  
Tech Biotech  
3DWHO < 1DUDLD(2012) 0HGLFK QDO SURSHUWLHV RI SOHXURWXV VSH  
World J Fungal Plant Biol  
&ORXJK 65 0X(2005) OHDQGH/ (QF\FORSHGLD RI WR[L86\$ DRJA QG HC  
7OLOL \$ (2007)The Application of dichloromethane and chloroform as reagents in organic synthesis; in:  
VROYHQWV DV UHJDJHQWV LQ RUJDQLF V86\$ KHVLV UHDFWLRQV DQ